Selectivity Switch in the Catalytic Functionalization of Nonprotected Carbohydrates: Selective Synthesis in the Presence of Anomeric and Structurally Similar Carbohydrates under Mild Conditions

Wataru Muramatsu* and Yuki Takemoto

Graduate School of bio[me](#page-8-0)dical Sciences, Nagasaki University, 1-14 Bunkyo-machi, Nagasaki, Nagasaki 852-8521, Japan

S Supporting Information

[AB](#page-8-0)STRACT: [A catalytic p](#page-8-0)rocess for the chemo- and regioselective functionalization of nonprotected carbohydrates has been developed. This novel process allows selective thiocarbonylation, acylation, and sulfonylation of a particular hydroxy group in a particular carbohydrate in the simultaneous presence of structurally similar carbohydrates such as anomers. In addition, the chemoselectivity can be switched by regulating only the length of the alkyl chain in the organotin catalyst.

ENTRODUCTION

Carbohydrates including oligosaccharides and glycoconjugates have been known to play crucial roles in various physiologic and pathologic events such as cell adhesion, fertilization, and cancer cell metastases.¹ With the aim to gain insight into the mechanism of these events and to develop new medicines, the development of efficie[nt](#page-9-0) routes and comprehensive procedures for the high purity synthesis of such carbohydrates is indispensable. In the past years, various methods for stereoand regioselective synthesis of carbohydrates have been developed.² However, these methods do not necessarily afford high stereo- and regioselectivity to all the substrates, and subsequen[t](#page-9-0) separation of anomers by simple methods such as silica gel column chromatography and recrystallization is generally very difficult as anomers often have similar polarity and solubility properties. In addition, special instrument techniques such as high-performance liquid chromatography (HPLC) are unsuitable for the fast separation of a large quantity of anomers. Therefore, effort-consuming derivatization of anomers with multistep protection−deprotection sequences has been often attempted for separation of anomers. To resolve such problems, enzymatic method for the separation of an anomeric mixture of carbohydrates and structurally similar carbohydrates has been developed in the last few decades.³ For example, Gotor and co-workers reported the separation method of an anomeric mixture of α - and β -D-nucleosides th[ro](#page-9-0)ugh regioselective lipase-catalyzed acylation.3d More recently, Prasad and co-workers reported the separation method of a mixture of furanosyl and pyranosyl nucleosi[de](#page-9-0)s through chemoand regioselective lipase-catalyzed acylation.^{3e} These catalyses are useful techniques for the separation with protection (acyl) group in a minimum number of steps. Ho[wev](#page-9-0)er, the resulting functionalized nucleosides and carbohydrates are usually not the ideal precursors for further functionalization. To the best of our knowledge, on the other hand, a nonenzymatic method has not been reported to date. Herein, we present research to facilitate chemical functionalization and separation of carbohydrates utilizing the high molecular recognition ability of organotin catalysts.

■ RESULTS AND DISCUSSION

In the last a few years, we have investigated a catalytic method for the regioselective introduction of useful functional groups into nonprotected carbohydrates.^{2j,4} In these results, we found that Me₂SnCl₂-catalyzed reaction at C(6)-OH of methyl β -Dglucopyranoside was much [fast](#page-9-0)er than Bu_2SnCl_2 (or Oc_2SnCl_2)-catalyzed reaction. On the other hand, Bu_2SnCl_2 (or Oc_2SnCl_2)-catalyzed reaction at C(2)-OH of methyl α -Dglucopyranoside was faster than $Me₂SnCl₂$ -catalyzed reaction. Therefore, we expect that a particular carbohydrate can be selectively functionalized (= extracted) among structurally similar carbohydrates by regulating the length of the alkyl chain in the organotin catalyst.

First of all, we investigated the chemo- and regioselective thiocarbonylation of methyl D-glucopyranosides with various organotin catalysts (10 mol %) in entries 1−16 in Table 1. Treatment of a mixture of methyl α -D-glucopyranoside and methyl β -D-glucopyranoside (1.0 equiv) with phenyl chl[or](#page-1-0)othionoformate (1.3 equiv) as an electrophile, 1,2,2,6,6 pentamethylpiperidine (PEMP, 1.5 equiv), and tetrabutylammonium iodide (TBAI, 10 mol %) in the absence of organotin catalysts in THF at 20 °C gave no products (entry 1). In the presence of $Me₂SnCl₂$, methyl β -D-glucopyranoside was selectively converted into methyl 6-O-phenoxythiocarbonyl-β-D-glucopyranoside 1b without formation of its regioisomers (entry 2; 93% yield and 79% chemoselectivity). On the other hand, $Bu₂SnCl₂$ favored selective formation of methyl 2-Ophenoxythiocarbonyl-α-D-glucopyranoside 1a without formation of its regioisomers (entry 3; 80% yield and 95% chemoselectivity). High selectivities to 1a were also afforded

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Table 1. Scope of a Catalyst in Chemo- and Regioselective Thiocarbonylation

 a ^aThe yield and ratio were calculated by ¹H NMR after isolation of mixture of 1a and 1b. b Oc = n-octyl. c Dd = n-dodecyl. d Without TBAI. ^e3,5-Lutidine was added instead of TBAI. ^f3,5-Diphenylpyridine was added instead of TBAI. ^gDMAP was added instead of TBAI. was added instead of TBAI. ⁸DMAP was added instead of TBAI.
^hTMAI was added instead of TBAI. ⁱTBABr was added instead of TBAI. ^j Under optimized conditions (see the Experimental Section).

with Oc_2SnCl_2 and Dd_2SnCl_2 (entries [5 and 6\), whereas](#page-4-0) t- Bu_2SnCl_2 and Ph_2SnCl_2 were completely ineffective (entries 4 and 7). Dimethyltin and dibutyltin derivatives were also employed as catalysts in the titled reaction. Unlike $Me₂SnO$ (entry 9), $Me₂SnBr₂$ and $Me₂SnS$ afforded the desired product 1b in 90−91% yields with 78−79% chemoselectivities (entries 8 and 10). On the other hand, organotin catalysts containing a butyl group furnished 1a with yields up to 66% and chemoselectivities up to >99% (entries 11−16). Next, we investigated a screening of additive reagent instead of TBAI. In the case of $Me₂SnCl₂$ without TBAI, the catalysis gave better yield and chemoselectivity (entry 17; 97% yield and 83% chemoselectivity). Addition of 3,5-lutidine (10 mol %) instead of TBAI improved the chemoselectivity (entry 18; 87% yield and 86% chemoselectivity). However, addition of the other pyridine derivatives such as DMAP or ammonium salts did not bring a superior result (entries 19 and 20). On the other hand, $Bu₂SnCl₂$ -catalyzed reaction without TBAI gave slightly lower yield (entry 21; 73% yield and 95% chemoselectivity). Additionally, addition of the other pyridine derivatives such as DMAP or ammonium salts also did not give better results (entries 22−24). Finally after a series of optimization studies, we found that the selective thiocarbonylation at C(6)-OH of methyl β -D-glucopyranoside in the presence of methyl α -Dglucopyranoside (1.0 equiv), $Me₂SnCl₂$ (10 mol %), 3,5lutidine (10 mol %), and PEMP (1.5 equiv) in THF at 20 °C with slow addition of phenyl chlorothionoformate (1.3 equiv) proceeded efficiently in 91% yield and 98% chemoselectivity without formation of its regioisomers (entry 25). On the other hand, selective thiocarbonylation at $C(2)$ -OH of methyl α -Dglucopyranoside in the presence of methyl β -D-glucopyranoside (1.0 equiv) , Bu₂SnCl₂ (10 mol %), TBAI (50 mol %),⁴ PEMP (1.3 equiv), and phenyl chlorothionoformate (1.3 equiv) in THF at 20 °C proceeded efficiently in >99% yield [an](#page-9-0)d 97% chemoselectivity without formation of its regioisomers (entry 26).

Once optimized conditions were identified, we next performed the Bu₂SnCl₂-catalyzed chemo- and regioselective functionalization with respect to both the substituent $R¹$ at $C(1)$ -position of D-glucopyranosides and several kinds of electrophiles (Table 2). Substrates with O-alkyl or O-aryl

10 OMe/Bz 11a, 11b >99:8 93:7 11 OMe/i-PrC(O) 12a, 12b 96:11 90:10 12 $OMe/(3,5-CF_3)$ -PhSO₂ 13a, 13b 97:14 87:13 ^aThe yield and ratio were calculated by ¹H NMR after isolation of mixture of 2–13a and 2–13b. $\frac{b}{b}PNP = p$ -nitrophenyl.

substituents $R¹$ at the C(1)-position were readily thiocarbonylated at C(2)-OH of α-D-glucopyranosides with 92−96% yields and 96−98% chemoselectivities (entries 1−3). Ethyl β-Dthioglucopyranoside,⁵ a useful glycosyl donor, was converted under the same reaction conditions to the corresponding thiocarbonate 5a w[it](#page-9-0)h 92% yield and 91% chemoselectivity (entry 4). The use of chlorothionoformates R^2 -Cl containing an electron-donating or electron-withdrawing group as an electrophile resulted in strong reactivities and high selectivities at C(2)-OH, whereas dimethylthiocarbamoyl chloride did not show reactivity for this reaction (entries 5−9). Remarkably, the protocol herein described can be applied for the chemo- and regioselective introduction of acyl protection groups as well as sulfonyl functionalities serving as a good leaving group for the subsequent nucleophilic substitution reactions (entries 10−12). In these catalytic reactions, the corresponding regioisomers were not observed.

On the other hand, the results for the $Me₂SnCl₂-catalyzed$ chemo- and regioselective functionalization with respect to both the substituent R^1 at $C(1)$ -position of D-glucopyranosides and several kinds of electrophiles \mathbb{R}^2 -Cl are shown in Table 3.

| entry | R^1/R^2 | products | y is in (a/b, %) | 1 uu v (a/b) |
|--|-------------------------------------|----------|-----------------------|-----------------|
| $\mathbf{1}$ | OOc/PhOC(S) | 2a, 2b | 12:85 | 12:88 |
| 2 | OPh/PhOC(S) | 3a, 3b | 14:68 | 17:83 |
| 3 | OPNP ^b /PhOC(S) | 4a, 4b | 18:68 | 21:79 |
| 4 | SEt/PhOC(S) | 5a, 5b | 5:79 | 6:94 |
| 5 | $OMe/(4-Tol)OC(S)$ | 6a, 6b | 6:91 | 6:94 |
| 6 | $OMe/(2-Naph)OC(S)$ | 7a, 7b | 3:80 | 4:96 |
| 7 | $OMe/(4-Cl-Ph)OC(S)$ | 8a, 8b | 4:82 | 4:96 |
| 8 | $OMe/(4-F-Ph)OC(S)$ | 9a, 9b | 6:90 | 6:94 |
| 9 | OMe/Me ₂ NC(S) | 10a, 10b | 0:0 | nd |
| 10 | OMe/Bz | 11a, 11b | 7:97 | 7:93 |
| 11 | $OMe/i\text{-}PrC(O)$ | 12a, 12b | <1:92 | <1:>99 |
| 12 | $OMe/(3,5-CF_3)$ -PhSO ₂ | 13a, 13b | 1:>99 | <1:>99 |
| ^{<i>a</i>The yield and ratio were calculated by ¹H NMR after isolation of} | | | | |

The yield and ratio were calculated by ¹H NMR after isolation of mixture of 2–13a and 2-13b. $\binom{b}{r}$ PNP = p-nitrophenyl.

In the screening of the $R¹$ substituent, selective thiocarbonylation at C(6)-OH of β -D-glucopyranosides proceeded in 68– 85% yields and 79−94% chemoselectivities without formation of its regioisomers (entries 1−4). Similarly, the catalytic thiocarbonylation with various chlorothionoformates R^2 -Cl as electrophiles proceeded in high yields and excellent selectivities (entries $5-8$). As in the case of Bu₂SnCl₂, dimethylthiocarbamoyl chloride was not effective under the reaction conditions of the present study (entry 9). Again, this protocol can be further applied for chemo- and regioselective acylation and sulfonylation with 92−>99% yields and 93−>99% chemoselectivities (entries 10−12).

The mechanism explaining the chemo- and regioselectivities in the catalytic functionalization of D-glucopyranosides can be rationalized as follows (Scheme 1a).⁶ Organotin catalysts coordinate reversibly to cis-1,2- or 1,3-diol moieties after moving freely among diol moieties. [As](#page-9-0) the coordination of metal ions may increase the acidity of hydroxy groups, even a weak base such as PEMP is sufficient to induce deprotonation of both hydroxy groups leading to five- or six-membered ring

Scheme 1. Plausible Mechanism of the Chemo- and Regioselective Functionalization

(a) Catalytic cycle in the selective functionalization.

(b) Negative intermediates with electronic and steric effects.

(c) Formation of the intermediates for promotion of reactivity and selectivity

intermediates A, B, or C whose subsequent irreversible functionalization proceeds much more rapidly at the less hindered hydroxy group. In the case of the use of organotin catalysts containing long alkyl groups such as Bu_2SnCl_2 and Oc_2SnCl_2 , the catalytic functionalization proceeds with high chemo- and regioselectivities at $C(2)$ -OH of α -D-glucopyranosides via kinetically stabilized intermediate B as the other intermediates A and C are unstabilized by the 1,3-diaxial interaction such as C′ between the alkyl group R and axial-H at $C(4)$ - and $C(6)$ -positions (Scheme 1b). Additionally, comparatively more reactive electrophiles R^2 -I, generated from R^1 -Cl and TBAI, accelerate the reaction (Scheme 1c).

In the case of the use of $Me₂SnCl₂$, the catalytic functionalization proceeds selectively at the less hindered hydroxy group, $C(6)$ -OH, of β -D-glucopyranosides through the intermediate C. Additionally, a bulky pyridinium intermediate, generated from R¹-Cl and 3,5-lutidine (Scheme 1c), yielded 6-O-functionalized- β -D-glucopyranosides with higher selectivities. On the other hand, the reaction at $C(2)$ -OH of α -Dglucopyranosides with the intermediate formation of B is considerably slower than that involving the formation of the intermediate C because of steric hindrance around C(2)-OH of α -D-glucopyranosides. Furthermore, we investigated a comparison of potential reactivities of C(6)-OH in α -D-Glc with C(6)-OH in β -D-Glc under standard acylation conditions (Figure 1). As a result, we could not find a dominant difference between the reactivities of C(6)-OH in β [-D](#page-3-0)-Glc and C(6)-OH in α -D-Glc. This shows that $Me₂SnCl₂$ clearly controls chemo-

Figure 1. Comparison of the reactivities of C(6)-OH in glucopyranosides.

selectivity under catalytic conditions, and we consequently suppose that nucleophilicity (reactivity) of $C(6)$ -OH in β -D-Glc is weakened because of a decline of the electron density of C(6)-OH in β -D-Glc in the intermediates A by a long-range stereoelectronic effect (anomeric effect) as shown A′ in Scheme 1b.^{7,8} However, it is still unclear.

As shown in Scheme 2, the protocol can be applied to the [ch](#page-2-0)[em](#page-9-0)o- and regioselective functionalization in the presence of structurally similar carbohydrates.⁹ In the case of thiocarbonylation with Bu_2SnCl_2 in the simultaneous presence of a pyranoside and furanoside co[nt](#page-9-0)aining cis-1,2-diol moieties (for example, α -D-Glc vs α -D-Ara), C(2)-OH of α -D-Glc was thiocarbonated with 97% yield and >99% chemoselectivity without formation of its regioisomers. Additionally, the catalytic sulfonylation afforded the corresponding sulfonate 15 in high yield with excellent selectivities (Scheme 2 (a)). $Bu₂SnCl₂$ catalyzed thiocarbonylation, acylation, and sulfonylation were effectively carried out in the presence of α -D-Glc and α -D-Man. Thus, in these reactions, α -D-Glc was converted to 2-Othiocarbonate 1a, 2-O-benzoate 11a, and 2-O-sulfonate 13a, respectively, in 90−>99% yields with 90−96% chemoselectivities (Scheme 2 (b)). Remarkably, these functionalizations selectively proceeded at $C(3)$ -OH of D-Gals in the presence of $D-Glcs$ (Scheme 2 (c) and (d)). Taking into consideration the above results, the most suitable 1,2-diol moieties in common carbohydrates that coordinate tightly and reversibly to Bu₂SnCl₂ are C(3)- and C(4)-OHs of D-Gals. We suppose that the most effective factor determining the coordination is the O−Sn−O binding angle. As shown in Scheme 2e, $Me₂SnCl₂-catalyzed functionalization in the simultaneous$ presence of $β$ -D-Glc and $β$ -D-Gal proceeded at C(6)-OH of β-D-Glc with 96−>99% yields and 94−96% chemoselectivities without formation of its regioisomers.

■ CONCLUSION

A catalytic process for the chemo- and regioselective functionalization of nonprotected carbohydrates in the presence of the stereoisomers has been developed. The present method with Bu_2SnCl_2 enables direct functionalization at $C(2)$ -OH of methyl α -D-glucopyranoside in the presence of methyl β -D-glucopyranoside in >99% yield with 97% chemoselectivity without formation of its regioisomers. On the other hand, Me₂SnCl₂ leads to direct functionalization at $C(6)$ -OH of methyl β -D-glucopyranoside in the presence of methyl α -Dglucopyranoside in 91% yield and 98% chemoselectivity without formation of its regioisomers. The chemo- and regioselectivities of functionalization are found to be intrinsic to the carbohydrates based on the affinity of metal ion with diol moieties in carbohydrates and the stereorelationship among hydroxy groups, respectively. In addition, the method can be applied to chemo- and regioselective functionalization in the presence of the structurally similar carbohydrates, thereby providing a novel and helpful strategy to efficiently separate inseparable carbohydrates. Additional efforts for extending these methods to more complex polyols including oligosaccharides and glycoconjugates, and mechanistic investigations are underway in our laboratories.

EXPERIMENTAL SECTION

General Methods. NMR spectra were recorded at 400, 100, and 376 MHz for ${}^{1}H$, ${}^{13}C$, and ${}^{19}F$ acquisitions, respectively. Chemical shifts are reported in ppm with a solvent resonance as an internal standard (¹ H NMR: tetramethylsilane, acetone, chloroform, or pyridine as internal standards, indicating 0, 2.05, 7.26, or 8.71, respectively; 13C NMR: acetone, methanol, chloroform, or pyridine as internal standards, indicating 0, 29.8, 77.0, or 135.5, respectively). ¹⁹F NMR: $CFCI₃$ as internal standard, indicating 0 ppm). Data is reported as follows: $s =$ singlet, $br =$ broad, $d =$ doublet, $t =$ triplet, $q =$ quartet, m = multiplet; coupling constants in Hz; integration. Assignments were based on analysis of coupling constants and COSY spectra. Mass spectra were recorded with fast atom bombardment (FAB) and a double-focusing magnetic sector mass analyzer for MS and HRMS measurements. All screenings of chemo- and regioselective functionalization were carried out in oven-dried screw-cap vials fitted with a septum.

General Procedure for the Chemo- and Regioselective Functionalization of Anomeric Mixture of Glucopyranoside Catalyzed by Bu_2SnCl_2 (Entry 26 in Table 1). After a mixture of methyl α -D-glucopyranoside (194.2 mg, 1.0 mmol), methyl β -Dglucopyranoside (194.2 mg, 1.0 mmol), and dibutyltin dichloride (30.4 mg, 0.10 mmol) in THF (8 mL) was stirr[ed](#page-1-0) in a vial at room temperature for 10 min, tetrabutylammonium iodide (184.7 mg, 0.50 mmol), phenyl chlorothionoformate (0.175 mL, 1.3 mmol), and 1,2,2,6,6-pentamethylpiperidine (0.235 mL, 1.3 mmol) were added to the suspension at 20 °C. After being stirred vigorously for 6 h at the same temperature, the reaction mixture was quenched with saturated aqueous NH4Cl and subsequently extracted with ethyl acetate. The organic layer was washed with water and brine, dried over $MgSO_4$, filtrated, and concentrated in vacuo (water bath temperature: $\langle 20 \degree C \rangle$. The residue was purified by $SiO₂$ column chromatography (hexane/ ethyl acetate = 3:1−0:1) to give a mixture of methyl 2-Ophenoxythiocarbonyl- α -D-glucopyranoside 1a (330.9 mg, >99%) and methyl 6-O-phenoxythiocarbonyl-β-D-glucopyranoside 1b (10.2 mg,

3%).
Methyl 2-O-Phenoxythiocarbonyl- α -D-glucopyranoside (1a, Entry 26, Table 1).^{4,10} Major product 1a: 330.9 mg, >99% yield. Minor product 1b: 10.2 mg, 3% yield. White solid: $R_f = 0.27$ (MeOH/ CHCl₃, 10:90); mp [130](#page-9-0)–132 °C; $[\alpha]_{D}^{26}$ = +96.4 (c 1.02, CH₃OH).
¹H NMP (400 MHz, C D N) δ 7.35–7.30 (m 2H PhH) 7.21–7.11 ¹[H](#page-1-0) NMR (400 MHz, C_5D_5N) δ 7.35–7.30 (m, 2H, PhH), 7.21–7.11 $(m, 3H, PhH)$, 5.88 (dd, J = 9.9, 3.4 Hz, 1H, H-2), 5.58 (d, J = 3.4 Hz, 1H, H-1), 4.86 (t, J = 9.2 Hz, 1H, H-3), 4.51−4.48 (m, 1H, H-6a), 4.42−4.25 (m, 3H, H-5 and H-6b), 3.42 (s, 3H, OCH₃); ¹³C NMR $(100 \text{ MHz}, \text{ C}_5\text{D}_5\text{N}) \delta$ 196.0, 154.0, 129.9 (2C), 126.8, 122.4 (2C), 96.8, 84.5, 74.3, 72.1, 71.8, 62.3, 54.9; IR (solid) 3323, 2900, 1489, 1269, 1221, 1204, 1043, 1016 cm⁻¹; MS (FAB) m/z (rel intensity) 331 (M + H⁺ , 20), 307 (25), 289 (15), 177 (5), 154 (100), 136 (70), 107 (20), 77 (20); HRMS (FAB) calcd for $C_{14}H_{19}O_7S$ $(M + H^+)$ 331.0846, found 331.0855. Anal. Calcd for C₁₄H₁₈O₇S: C, 50.90; H, 5.49. Found: C, 50.61; H, 5.33.

Octyl 2-O-Phenoxythiocarbonyl- α -D-glucopyranoside (2a, Entry 1, Table 2). 4 Major product 2a: 409.4 mg, 96% yield. Minor product **2b**: 17.1 mg, 4% yield. White solid: $R_f = 0.40$ (MeOH/CHCl₃, 10:90); mp 100−102 °C; [α]²⁶_D = +125.5 (α 1.00, CH₃OH); ¹H NMR (400 MHz, C_5D_5N) δ 7.60–7.37 (m, 2H, PhH), 7.27–7.18 (m, 3H, PhH), 5.89 (dd, J = 9.8, 3.7 Hz, 1H, H-2), 5.78 (d, J = 3.7 Hz, 1H, H-1), 5.22 (br s, 3H, OH), 4.95−4.90 (m, 1H, H-3), 4.58−4.54 (m, 1H, H-6a), 4.45−4.38 (m, 3H, H-4, H-5 and H-6b), 4.00−3.93 (m, 1H, $OCH_2(CH_2)_6CH_3$, 3.59–3.52 (m, 1H, $OCH_2(CH_2)_6CH_3$), 1.70– 1.50 (m, 2H, $OCH_2CH_2(CH_2)_{5}CH_3$), 1.45−1.25 (m, 2H, $OCH_2CH_2CH_2CH_2(CH_2)_4CH_3$, 1.25 – 1.00 (m, 8H, $OCH_2CH_2CH_2(CH_2)_{4}CH_3$, 0.82 (t, J = 6.8 Hz, 3H, $OCH_2(CH_2)_6CH_3)$; ¹³C NMR (100 MHz, C₅D₅N) δ 195.9, 153.9, 129.9 (2C), 126.8, 122.3 (2C), 95.5, 84.4, 74.3, 72.0, 71.8, 68.1, 62.3, 31.9, 29.7, 29.5, 29.4, 26.3, 22.8, 14.1; IR (solid) 3447, 2926, 1489, 1342, 1283, 1204, 1049, 1020 cm⁻¹; MS (FAB) m/z (rel intensity) 429 (M + H+ , 30), 307 (25), 299 (30), 281 (30), 275 (15), 145 (100), 77 (40), 57 (45); HRMS (FAB) calcd for $C_{21}H_{33}O_7S$ $(M + H^+)$ 429.1942, found 429.1960. Anal. Calcd for $C_{21}H_{32}O_7S$: C, 58.86; H, 7.53. Found: C, 58.65; H, 7.43.

Phenyl 2-O-Phenoxythiocarbonyl-α-D-glucopyranoside (3a, Entry 2, Table 2).⁴ Major product 3a: 374.6 mg, 92% yield. Minor product 3b: 7.6 mg, 2% yield. White solid: $R_f = 0.35$ (MeOH/CHCl₃, 10:90); mp [20](#page-9-0)4–206 °C; [α]²⁶_D = +157.3 (c 1.00, CH₃OH); ¹H NMR (400 MHz, C_5D_5N) δ 8.20 (br s, 1H, OH), 7.84 (br s, 1H, OH), 7.50−7.26 (m, 6H, PhH), 7.25−7.17 (m, 1H, PhH), 7.15−7.00 (m, 3H, PhH), 6.49 (d, J = 3.4 Hz, 1H, H-1), 6.49 (br s, 1H, OH), 6.04 (dd, J = 10.0, 3.4 Hz, 1H, H-2), 5.10 (t, J = 9.2 Hz, 1H, H-3), 4.60– 4.30 (m, 4H, H-4, H-5 and H-6); ¹³C NMR (100 MHz, C₅D₅N) δ 195.9, 157.6, 153.9, 130.0 (2C), 129.9 (2C), 126.9, 123.1, 122.3 (2C), 117.7 (2C), 95.3, 83.8, 75.3, 72.0, 71.4, 61.9; IR (solid) 3412, 2941, 1492, 1261, 1217, 1176, 1099, 1022 cm⁻¹; MS (FAB) m/z (rel intensity) 393 (M + H⁺, 25), 354 (5), 307 (25), 289 (15), 154 (100), 136 (70), 107 (20), 77 (20); HRMS (FAB) calcd for $C_{19}H_{21}O_7S$ (M + H⁺) 393.1003, found 393.1015. Anal. Calcd for C₁₉H₂₀O₇S: C, 58.15; H, 5.14. Found: C, 57.87; H, 5.26.

4-Nitrophenyl 2-O-Phenoxythiocarbonyl-α-D-glucopyranoside (4a, Entry 3, Table 2).⁴ Major product 4a: 404.1 mg, 92% yield. Minor product 4b: 8.2 mg, 2% yield. White solid: $R_f = 0.35$ (MeOH/ CHCl₃, 10:90); mp 180–182 °C; $[\alpha]_{2D}^{2D} = +221.6$ (c 1.00, CH₃OH);
¹H NMB (400 MHz, C D N) δ 8 18 (d I – 9.0 Hz, 2H, 4 NO, PbH) ¹H NMR (400 MHz, C_5D_5N) δ 8.18 (d, J = 9.0 Hz, 2H, 4-NO₂-PhH), 7.37-7.33 (m, 4H, 4-NO₂-PhH and PhH), 7.25-7.14 (m, 3H, PhH), 6.61 (d, $J = 3.2$ Hz, 1H, $H-1$), 6.12 (dd, $J = 9.9$, 3.2 Hz, 1H, $H-2$), 5.31 $(br s, 3H, OH), 5.09 (t, J = 9.3 Hz, 1H, H₋₃), 4.57–4.34 (m, 4H, H₋₄,$ H-5 and H-6); ¹³C NMR (100 MHz, C₅D₅N) δ 195.9, 161.9, 154.0, 150.0, 130.0 (2C), 127.0, 126.0 (2C), 122.3 (2C), 117.2 (2C), 95.0, 83.3, 76.0, 71.9, 71.2, 61.8; IR (solid) 3385, 2928, 2361, 1593, 1516, 1344, 1244, 1213, 1020 cm⁻¹; MS (FAB) m/z (rel intensity) 438 (M + H+ , 5), 307 (35), 289 (20), 154 (100), 136 (65), 107 (20), 77 (20); HRMS (FAB) Calcd for $C_{19}H_{20}NO_9S$ $(M + H⁺)$ 438.0853, found 438.0867. Anal. Calcd for C₁₉H₁₉NO₉S: C, 52.17; H, 4.38; N, 3.20. Found: C, 51.77; H, 4.23; N, 2.98.

Ethyl 2-O-Phenoxythiocarbonyl-α-D-thioglucopyranoside (5a, Entry 4, Table 2).⁴ Major product 5a: 331.1 mg, 92% yield. Minor product 5b: 32.8 mg, 9% yield. White solid: R_f = 0.32 (MeOH/CHCl₃, 10:90); mp [16](#page-9-0)5−167 °C; [α]²⁸_D = +193.4 (c 1.09, CH₃OH); ¹H NMR (400 MHz, C_5D_5N) δ 8.14 (br s, 1H, OH), 7.76 (br s, 1H, OH), 7.37 $(t, J = 7.8 \text{ Hz}, 2H, PhH), 7.30–7.10 \text{ (m, 3H)}, 6.40 \text{ (d, } J = 5.6 \text{ Hz}, 1H,$ H-1), 6.00 (dd, J = 9.9, 5.6 Hz, 1H, H-2), 5.11 (br s, 1H, OH), 4.78 (t, J = 9.5 Hz, 1H, H-3), 4.80−4.70 (m, 1H, H-5), 4.54 (dd, J = 12.0, 2.1 Hz, 1H, H-6a), 4.45 (dd, J = 12.0, 5.1 Hz, 1H, H-6b), 4.39 (t, J = 9.4 Hz, 1H, H-4), 2.78–2.62 (m, 2H, SCH₂CH₃), 1.23 (t, J = 7.5 Hz, 3H, SCH_2CH_3); ¹³C NMR (100 MHz, C₅D₅N) δ 195.4, 153.9, 129.9 (2C), 126.9, 122.3 (2C), 83.7, 83.1, 74.8, 72.9, 71.8, 62.2, 24.3, 15.0; IR (solid) 3356, 2918, 1487, 1288, 1215, 1103, 1043, 1015 cm[−]¹ ; MS (FAB) m/z (rel intensity) 361 (M + H⁺, 5), 307 (25), 289 (15), 154 (100), 136 (70), 107 (20), 77 (20); HRMS (FAB) calcd for $C_{15}H_{21}O_6S_2$ (M + H⁺) 361.0774, found 361.0797. Anal. Calcd for $C_{15}H_{20}O_6S_2$: C, 49.98; H, 5.59. Found: C, 49.73; H, 5.70.

Methyl 2-O-(4-Tolyloxy)thiocarbonyl- α -D-glucopyranoside (6a, Entry 5, Table 2).⁴ Major product 6a: 327.4 mg, 95% yield. Minor product 6b: 20.9 mg, 6% yield. White solid: $R_f = 0.35$ (MeOH/CHCl₃,

10:90); mp 148–150 °C; $[\alpha]_{D}^{26}$ = +8.4 (c 1.00, CH₃OH); ¹H NMR $(400 \text{ MHz}, \text{C}_5\text{D}_5\text{N}) \delta$ 7.06 (d, J = 8.1 Hz, 2H, 4-CH₃-PhH), 7.01 (d, J $= 8.1$ Hz, 2H, 4-CH₃-PhH), 5.92 (dd, J = 10.0, 3.7 Hz, 1H, H-2), 5.62 $(d, J = 3.7 \text{ Hz}, 1H, H-1), 4.89 \text{ (dd, } J = 10.0, 8.7 \text{ Hz}, 1H, H-3), 4.52$ $(dd, J = 12.0, 2.2 Hz, 1H, H-6a), 4.44-4.37 (m, 2H, H-4 and H-6b),$ 4.33−4.28 (m, 1H, ^H-5), 3.45 (s, 3H, OCH3), 2.12 (s, 3H, 4-CH3-Ph); 13C NMR (100 MHz, C5D5N) ^δ 196.2, 151.8, 136.4, 130.3 (2C), 121.9 (2C), 96.7, 84.4, 74.2, 72.1, 71.8, 62.2, 54.8, 20.6; IR (solid) 3385, 2907, 1504, 1271, 1219, 1192, 1038, 1015 cm⁻¹; MS (FAB) m/z (rel intensity) 345 (M + H⁺, 15), 291 (40), 235 (15), 177 (20), 145 (80), 99 (100), 91 (95), 71 (85); HRMS (FAB) calcd for $C_{15}H_{21}O_7S$ $(M + H⁺)$ 345.1014, found 345.1003. Anal. Calcd for $C_{15}H_{20}O_7S$: C, 52.31; H, 5.85. Found: C, 52.52; H, 5.72.

Methyl 2-O-(2-Naphthoxy)thiocarbonyl-α-D-glucopyranoside (7a, Entry 6, Table 2).⁴ Major product 7a: 381.4 mg, >99% yield. Minor product 7b: 7.8 mg, 2% yield. White solid: $R_f = 0.35$ (MeOH/ CHCl₃, 10:90); mp 14[2](#page-9-0)–144 °C; $\left[\alpha\right]_{D}^{19} = +88.6$ (c 1.02, CH₃OH);
¹H NMR (400 MHz C D N) δ 7.89–7.80 (m 3H NaphH) 7.59– ¹H NMR (400 MHz[,](#page-1-0) C_5D_5N C_5D_5N) δ 7.89–7.80 (m, 3H, NaphH), 7.59– 7.56 (m, 1H, NaphH), 7.50−7.44 (m, 2H, NaphH), 7.32 (dd, J = 8.9, 2.3 Hz, 1H, NaphH), 6.27 (br s, 3H, OH), 5.95 (dd, J = 9.9, 3.7 Hz, 1H, H -2), 5.66 (d, $J = 3.7$ Hz, 1H, H -1), 4.93 (t, $J = 9.3$ Hz, 1H, H -3), 4.54 (dd, J = 11.8, 2.1 Hz, 1H, H-6a), 4.46−4.40 (m, 2H, H-4 and H-6b), 4.35−4.30 (m, 1H, H-5), 3.50 (s, 3H, OCH3); 13C NMR (100 MHz, C₅D₅N) δ 196.0, 151.5, 134.1, 132.1, 129.9, 128.2, 128.1, 127.1, 126.5, 121.8, 119.4, 96.8, 84.6, 74.3, 72.1, 71.8, 62.2, 54.9; IR (solid) 3348, 2907, 1508, 1265, 1227, 1193, 1030, 1015 cm⁻¹; MS (FAB) m/z (rel intensity) 381 (M + H⁺, 20), 307 (25), 242 (55), 154 (100), 136 (70), 117 (20), 77 (20); HRMS (FAB) calcd for $C_{18}H_{21}O_7S (M + H^+)$ 381.1003, found 381.0999. Anal. Calcd for $C_{18}H_{20}O_7S$: C, 56.83; H, 5.30. Found: C, 56.54; H, 5.24.

Methyl 2-O-(4-Chlorophenoxy)thiocarbonyl-α-D-glucopyranoside (8a, Entry 7, Table 2).⁴ Major product 8a: 364.2 mg, >99% yield. Minor product 8b: 19.2 mg, 5% yield. White solid: $R_f = 0.30$ (MeOH/CHCl₃, [1](#page-9-0)0:90); mp 133–134 °C; $[\alpha]_{\text{D}}^{19}$ = +100.5 (c 1.05, CH₃OH); ¹H NMR (400 [MH](#page-1-0)z, C₅D₅N) δ 7.37 (d, J = 8.8 Hz, 2H, 4-Cl-PhH), 7.24 (br s, 3H, OH), 7.08 (d, $J = 8.8$ Hz, 2H, 4-Cl-PhH), 5.87 (dd, J = 9.8, 3.7 Hz, 1H, H-2), 5.59 (d, J = 3.7 Hz, 1H, H-1), 4.87 $(t, J = 9.3 \text{ Hz}, 1H, H-3), 4.52 \text{ (dd, } J = 11.7, 2.2 \text{ Hz}, 1H, H-6a), 4.42 \text{ (t, }$ $J = 5.1$ Hz, 1H, H-4), 4.38 (d, $J = 11.7$ Hz, 1H, H-6b), 4.32–4.27 (m, 1H, H-5), 3.45 (s, 3H, OCH₃); ¹³C NMR (100 MHz, C₅D₅N) δ 195.5, 152.3, 131.9, 129.9 (2C), 124.0 (2C), 96.6, 84.6, 74.2, 72.0, 71.7, 62.2, 54.8; IR (solid) 3443, 2936, 1487, 1281, 1206, 1146, 1042, 1013 cm⁻¹; MS (FAB) m/z (rel intensity) 365 (M + H⁺, 10), 307 (25), 289 (10), 242 (10), 154 (100), 136 (75), 107 (25); HRMS (FAB) calcd for $C_{14}H_{18}ClO_7S$ $(M + H^+)$ 365.0456, found 365.0475. Anal. Calcd for C14H17ClO7S: C, 46.09; H, 4.70. Found: C, 46.05; H, 4.60.

Methyl 2-O-(4-Fluorophenoxy)thiocarbonyl-α-D-glucopyranoside (9a, Entry 8, Table 2).⁴ Major product 9a: 360.5 mg, >99% yield. Minor product 9b: 11.1 mg, 3% yield. White solid: $R_f = 0.30$ (MeOH/ CHCl₃, 10:90); mp 125–127 °C; [α]²⁵_D = +0.6 (*c* 1.00, CH₃OH); ¹H NMR (400 MHz, C₅D₅N) δ 7.14–7.06 (m, 4H, 4-F-Ph*H*), 5.86 (dd, J $= 9.9, 3.7$ Hz, 1H, H-2), 5.57 (d, J = 3.7 Hz, 1H, H-1), 5.23 (br s, 3H, OH), 4.85 (t, J = 9.2 Hz, 1H, H-3), 4.50 (dd, J = 11.8, 2.1 Hz, 1H, H-6a), 4.42−4.34 (m, 2H, H-4 and H-6b), 4.30−4.25 (m, 1H, H-5), 3.43 (s, 3H, OCH₃); ¹³C NMR (100 MHz, C₅D₅N) δ 196.0, 160.9 (d, J = 244.1 Hz), 150.1, 124.1 (d, J = 8.3 Hz, 2C), 116.4 (d, J = 24.0 Hz, 2C), 96.7, 84.6, 74.3, 72.1, 71.8, 62.2, 54.9; IR (solid) 3368, 2936, 1503, 1271, 1210, 1190, 1042, 1013 cm⁻¹; MS (FAB) m/z (rel intensity) 349 (M + H⁺ , 5), 281 (10), 207 (10), 154 (35), 136 (40), 107 (35), 77 (40), 55 (100); HRMS (FAB) Calcd for $C_{14}H_{18}FO_7S$ $(M + H^+)$ 349.0752, found 349.0741. Anal. Calcd for $C_{14}H_{17}FO_7S$: C, 48.27; H, 4.92. Found: C, 48.36; H, 4.98.

Methyl 2-O-Benzoyl- α -D-glucopyranoside (11a, Entry 10, Table 2).^{11,12} Major product 11a: 303.3 mg, >99% yield. Minor product 11b: 22.8 mg, 8% yield. White solid: $R_f = 0.47$ (MeOH/CHCl₃, 10:90); mp 16[8](#page-9-0)−[17](#page-9-0)0 °C (lit.¹² mp 174−175 °C); $[\alpha]_{\text{D}}^{19} = +151.7$ (c 1.03, [C](#page-1-0)H₃OH) [lit.^{11a} [α]²⁵_D = +156.0 (c 2.09, EtOAc)]; ¹H NMR (400 MHz, $(CD_3)_2CO$) δ 8.07 (dd, J = 8.0, 1.2 Hz, 2H, PhH), 7.66 (t, J = 8.0 [H](#page-9-0)z, 1H, PhH), 7.53 (t, J = 8.0 Hz, 2H, PhH), 4.96 (d, J = 3.7 Hz, 1H, H-1), 4.83 (dd, J = 9.8, 3.7 Hz, 1H, H-2), 4.63 (br s, 1H, OH),

4.42 (br s, 1H, OH), 4.05 (t, J = 9.3 Hz, 1H, H-3), 3.90−3.80 (m, 1H, H-6a), 3.80−3.60 (m, 3H, H-5, H-6b and OH), 3.60−3.50 (t, J = 9.3 Hz, 1H, H-4), 3.35 (s, 3H, OCH₃); ¹³C NMR (100 MHz, $(CD_3)_2CO$) δ 166.6, 134.0, 131.1, 130.4 (2C), 129.3 (2C), 97.9, 75.2, 73.2, 72.2, 71.9, 62.5, 55.1; IR (solid) 3347, 2920, 1705, 1333, 1294, 1256, 1119, 1030 cm⁻¹; MS (FAB) m/z (rel intensity) 299 (M + H⁺, 20), 267 (20), 154 (100), 136 (70), 107 (20), 105 (25), 77 (15); HRMS (FAB) calcd for $C_{14}H_{19}O_7$ $(M + H^+)$ 299.1125, found 299.1159. Anal. Calcd for $C_{14}H_{18}O_7$: C, 56.37; H, 6.08. Found: C, 56.49; H, 6.05.

Methyl 2-O-Isobutyryl- α -D-glucopyranoside (12a, Entry 11, Table 2). Major product 12a: 253.7 mg, 96% yield. Minor product 12b: 29.0 mg, 11% yield. White solid: $R_f = 0.35$ (MeOH/CHCl₃, 10:90); mp 92−93 °C; [α]²⁵_D = +146.7 (c['] 1.00, CH₃OH); ¹H NMR (400 MHz, C_5D_5N C_5D_5N) δ 5.37 (dd, J = 10.0, 3.7 Hz, 1H, H-2), 5.29 (d, J = 3.7 Hz, 1H, H-1), 4.99 (br s, 2H, OH), 4.75−4.65 (m, 1H, H-3), 4.50 (d, J = 11.7 Hz, 1H, H-6a), 4.36 (d, J = 11.7 Hz, 1H, H-6b), 4.30−4.20 (m, 2H, H-4 and H-5), 3.43 (s, 3H, OCH₃), 2.64 (sept, $J = 7.0$ Hz, 1H, OCOCH(CH₃)₂), 1.18 (d, J = 7.0 Hz, 3H, OCOCH(CH₃)₂), 1.16 (d, $J = 7.0$ Hz, 3H, OCOCH(CH₃)₂); ¹³C NMR (100 MHz, CDCl₃) δ 177.4, 97.2, 73.0, 71.6, 71.0, 70.0, 61.3, 55.3, 33.8, 19.0, 18.8; IR (solid) 3296, 2916, 1715, 1196, 1155, 1107, 1038, 1009 cm⁻¹; MS (FAB) *m/z* (rel intensity) 265 (M + H⁺, 40), 233 (80), 154 (100), 137 (70), 136 (70), 107 (20), 89 (15), 77 (15); HRMS (FAB) calcd for $C_{11}H_{21}O_7$ $(M + H⁺)$ 265.1282, found 265.1319. Anal. Calcd for C₁₁H₂₀O₇: C, 49.99; H, 7.63. Found: C, 49.93; H, 7.85.

Methyl 2-O-[3,5-Bis(trifluoromethyl)benzenesulfonyl]-α-D-glucopyranoside (13a, Entry 12, Table 2^{2j} Major product 13a: 455.9 mg, 97% yield. Minor product 13b: 68.1 mg, 14% yield. White solid: R_f = 0.31 (MeOH/CHCl₃, 10:90); mp 1[08](#page-9-0)–109 °C; $[\alpha]_{D}^{17}$ = +79.5 (c 1.29, CH₃OH); ¹H NMR (400 MHz, $(CD_3)_2CO$) δ 8.55 (s, 2H, 3,5- $CF_3\text{-}PhH$), 8.47 (s, 1H, 3,5-CF₃-PhH), 4.88 (d, J = 3.7 Hz, 1H, H-1), 4.66 (br s, 1H, OH), 4.45 (br s, 1H, OH), 4.37 (dd, J = 9.8, 3.7 Hz, 1H, H-2), 3.79 (t, J = 9.3 Hz, 1H, H-3), 3.85−3.72 (m, 2H, H-6a and OH), 3.65 (dd, J = 11.7, 5.1 Hz, 1H, H-6b), 3.54–3.50 (m, 1H, H-5), 3.36 (t, $J = 9.3$ Hz, 1H, H-4), 3.36 (s, 3H, OCH₃); ¹³C NMR (100 MHz, $(CD_3)_2CO$) δ 140.4, 133.0 (q, J = 34.8 Hz, 2C), 129.7 (q, J = 3.3 Hz, 2C), 128.4 (sept, J = 3.3 Hz), 123.6 (q, J = 272.3 Hz, 2C), 98.2, 82.5, 73.1, 71.8, 71.6, 62.1, 55.2; 19F NMR (376 MHz, (CD₃)₂CO) δ –62.4 (s, 6F); IR (solid) 3377, 2932, 1381, 1362, 1279, 1179, 1134, 974 cm⁻¹; MS (FAB) m/z (rel intensity) 471 (M + H+ , 10), 421 (35), 361 (30), 277 (30), 213 (35), 154 (25), 145 (100), 127 (70); HRMS (FAB) calcd for $C_{15}H_{17}F_6O_8S$ (M + H⁺) 471.0543, found 471.0559. Anal. Calcd for $C_{15}H_{16}F_6O_8S$: C, 38.30; H, 3.43. Found: C, 38.29; H, 3.14.

General Procedure for the Chemo- and Regioselective Functionalization of Anomeric Mixture of Glucopyranoside **Catalyzed by Me₂SnCl₂ (Entry 25, Table 1).** After a mixture of methyl α -D-glucopyranoside (194.2 mg, 1.0 mmol), methyl β -Dglucopyranoside (194.2 mg, 1.0 mmol), and dimethyltin dichloride (22.0 mg, 0.10 mmol) in THF (6 mL) was s[tir](#page-1-0)red in a vial at room temperature for 10 min, 3,5-lutidine (0.0114 mL, 0.10 mmol) and 1,2,2,6,6-pentamethylpiperidine (0.271 mL, 1.5 mmol) were added to the suspension at 20 °C. Then, phenyl chlorothionoformate (0.175 mL, 1.3 mmol) in THF (2 mL) was flowed over 2 h at 20 °C. After being stirred vigorously for 6 h at 20 °C, the reaction mixture was quenched with saturated aqueous NH₄Cl and subsequently extracted with ethyl acetate. The organic layer was washed with water and brine, dried over MgSO4, filtrated, and concentrated in vacuo (water bath temperature: <20 °C). The residue was purified by SiO_2 column chromatography (hexane/ethyl acetate = 3:1−0:1) to give a mixture of methyl 2-O-phenoxythiocarbonyl- α -D-glucopyranoside 1a (6.2 mg, 2%) and methyl 6-O-phenoxythiocarbonyl- β -D-glucopyranoside 1b (301.7 mg, 91%).

Methyl 6-O-Phenoxythiocarbonyl-β-p-glucopyranoside (**1b,**
Entry 25, Table 1).^{4,10} Major product 1b: 301.7 mg, 91% yield. Minor product 1a: 6.2 mg, 2% yield. White solid: $R_f = 0.27$ (MeOH/ CHCl₃, 10:90); mp [143](#page-9-0)–144 °C; [α]²⁵_D = -14.0 (c 1.00, CH₃OH);
¹H NMR (400 MHz, C D N) δ 7.38 (t I - 7.8 Hz, 2H PbH) 7.25– ¹H NMR (400 M[Hz](#page-1-0), C₅D₅N) δ 7.38 (t, J = 7.8 Hz, 2H, PhH), 7.25− 7.18 (m, 3H, PhH), 5.41 (d, $J = 11.5$ Hz, 1H, H-6a), 5.20 (d, $J = 11.5$ Hz, 1H, H-6b), 4.76 (d, J = 7.6 Hz, 1H, H-1), 4.30−4.10 (m, 3H, H-3, H-4 and H-5), 4.05 (t, J = 8.3 Hz, 1H, H-2), 3.63 (s, 3H, OCH₃); ¹³C NMR (100 MHz, C_5D_5N) δ 195.6, 153.9, 129.9 (2C), 126.8, 122.3 (2C), 105.6, 78.1, 74.8, 74.6, 71.1, 56.8; IR (solid) 3424, 2920, 1489, 1263, 1207, 1159, 1098, 1016 cm⁻¹; MS (FAB) m/z (rel intensity) 331 (M + H⁺ , 30), 307 (20), 289 (15), 177 (5), 154 (100), 136 (70), 107 (20), 77 (25); HRMS (FAB) calcd for $C_{14}H_{19}O_7S$ $(M + H^+)$ 331.0846, found 331.0845. Anal. Calcd for C₁₄H₁₈O₇S: C, 50.90; H, 5.49. Found: C, 50.83; H, 5.37.

Octyl 6-O-Phenoxythiocarbonyl-β-D-glucopyranoside (2b, Entry 1, Table 3). Major product 2b: 364.4 mg, 85% yield. Minor product 2a: 49.7 mg, 12% yield. White solid: $R_f = 0.40$ (MeOH/CHCl₃, 10:90); mp 85–86 °C; $[\alpha]^{24}$ _D = −12.3 (c 1.03, CH₃OH); ¹H NMR (400 M[Hz,](#page-2-0) C_5D_5N) δ 7.67 (br s, 1H, OH), 7.55–7.30 (m, 2H, PhH), 7.41 (br s, 1H, OH), 7.30−7.15 (m, 3H, PhH), 5.44 (d, J = 11.2 Hz, 1H, H-6a), 5.22 (dd, J = 11.2, 5.0 Hz, 1H, H-6b), 4.99 (br s, 1H, OH), 4.88 (d, J = 7.8 Hz, 1H, H-1), 4.30−4.00 (m, 5H, H-2, H-3, H-4, H-5 and $OCH_2(CH_2)_6CH_3$, 3.71 (dt, J = 16.1, 6.8 Hz, 1H, $OCH_2(CH_2)_6CH_3$, 1.75−1.60 (m, 2H, $OCH_2CH_2(CH_2)_5CH_3$), 1.40−1.25 (m, 2H, OCH₂CH₂CH₂(CH₂)₄CH₃), 1.25−1.05 (m, 8H, $OCH_2CH_2CH_2(CH_2) _4CH_3$, 0.82 (t, J = 7.0 Hz, 3H, OCH₂(CH₂)₆CH₃); ¹³C NMR (100 MHz, C₅D₅N) δ 195.7, 154.0, 129.9 (2C), 126.8, 122.4 (2C), 104.8, 78.2, 75.0, 74.8, 74.7, 71.3, 70.0, 31.9, 30.2, 29.6, 29.4, 26.3, 22.8, 14.2; IR (solid) 3414, 2924, 1491, 1375, 1290, 1200, 1082, 1020 cm⁻¹; MS (FAB) m/z (rel intensity) 429 (M + H⁺ , 20), 299 (90), 281 (25), 154 (40), 145 (100), 127 (65), 85 (95), 57 (65); HRMS (FAB) calcd for $C_{21}H_{33}O_7S$ $(M + H^+)$ 429.1942, found 429.1960. Anal. Calcd for $C_{21}H_{32}O_7S$: C, 58.86; H, 7.53. Found: C, 58.86; H, 7.78.

Phenyl 6-O-Phenoxythiocarbonyl-β-D-glucopyranoside (3b, Entry 2, Table 3). Major product 3b: 266.7 mg, 68% yield. Minor product 3a: 54.6 mg, 14% yield. White solid: $R_f = 0.35$ (MeOH/CHCl₃, 10:90); mp 140−141 °C; $[\alpha]_{\text{D}}^{26}$ = −65.5 (c 1.06, CH₃OH); ¹H NMR (400 M[Hz](#page-2-0), C_5D_5N) δ 8.00–7.80 (m, 2H, PhH and OH), 7.65 (br s, 1H, OH), 7.50−7.30 (m, 5H, PhH), 7.30−7.10 (m, 3H, PhH), 7.04 (t, $J = 7.1$ Hz, 1H, PhH), 6.60 (d, $J = 7.1$ Hz, 1H, H-1), 5.39 (d, $J = 11.5$ Hz, 1H, H-6a), 5.25 (dd, J = 11.5, 5.9 Hz, 1H, H-6b), 5.02 (br s, 1H, OH), 4.50–4.20 (m, 4H, H-2, H-3, H-4 and H-5); ¹³C NMR (100 MHz, C_5D_5N) δ 195.6, 158.6, 154.0, 129.9 (4C), 126.8, 122.5, 122.4 (2C), 117.0 (2), 102.2, 78.1, 74.8, 74.7, 74.4, 71.0; IR (solid) 3389, 2887, 1489, 1279, 1221, 1196, 1070, 1016 cm[−]¹ ; MS (FAB) m/z (rel intensity) 393 (M + H⁺, 10), 307 (20), 299 (10), 289 (15), 154 (100), 136 (70), 107 (20), 77 (20); HRMS (FAB) Calcd for C₁₉H₂₁O₇S (M + H⁺) 393.1003, found 393.1015. Anal. Calcd for C₁₉H₂₀O₇S: C, 58.15; H, 5.14. Found: C, 57.89; H, 5.31.

4-Nitrophenyl 6-O-Phenoxythiocarbonyl-β-D-glucopyranoside (4b, Entry 3, Table 3). Major product 4b: 299.3 mg, 68% yield. Minor product 4a: 79.5 mg, 18% yield. White solid: $R_f = 0.35$ (MeOH/CHCl₃, 10:90); mp 66–69 °C; $[\alpha]^{25}$ _D = -111.4 (c 1.05, CH₃OH); ¹H NMR ([40](#page-2-0)0 MHz, C₅D₅N) δ 8.24 (d, J = 9.0 Hz, 2H, 4-NO2-PhH), 8.09 (br s, 1H, OH), 7.58 (br s, 1H, OH), 7.45−7.27 (m, 4H, 4-NO2-PhH and PhH), 7.27−7.10 (m, 3H, PhH), 5.81 (d, J = 7.3 Hz, 1H, H-1), 5.44 (dd, J = 11.5 Hz, H-6a), 5.26 (dd, J = 11.5, 5.8 Hz, 1H, H-6b), 5.03 (br s, 1H, OH), 4.60−4.47 (m, 1H, H-5), 4.47−4.20 (m, 3H, H-2, H-3 and H-4); ¹³C NMR (100 MHz, C₅D₅N) δ 195.6, 163.0, 154.0, 142.7, 130.0 (2C), 126.8, 126.1 (2C), 122.3 (2C), 116.8 (2C), 101.5, 78.0, 75.1, 74.5, 74.3, 70.8; IR (solid) 3366, 2895, 1591, 1514, 1344, 1240, 1198, 1016 cm⁻¹; MS (FAB) m/z (rel intensity) 438 (M + H⁺ , 5), 307 (30), 289 (15), 154 (100), 136 (75), 107 (20), 77 (15); HRMS (FAB) Calcd for $C_{19}H_{20}NO_9S (M + H⁺)$ 438.0853, found 438.0841. Anal. Calcd for $C_{19}H_{19}NO_9S$: C, 51.89; H, 4.38; N, 3.08. Found: C, 51.77; H, 4.23; N, 2.98.

Ethyl 6-O-Phenoxythiocarbonyl-β-p-thioglucopyranoside (5b, Entry 4, Table 3). Major product 5b: 284.2 mg, 79% yield. Minor product 5a: 18.1 mg, 5% yield. White solid: $R_f = 0.32$ (MeOH/CHCl₃, 10:90); mp 89–90 °C; $[\alpha]_{D}^{26} = -31.3$ (c 1.01, CH₃OH); ¹H NMR (400 MHz, C₅D₅N) δ 7.38 (t, J = 7.8 Hz, 2H, PhH), 7.30–7.10 (m, 3H), 5.41 (dd, J = 11.5, 1.5 Hz, 1H, H-6a), 5.17 (dd, J = 11.5, 5.9 Hz, 1H, H-6b), 5.02 (d, J = 9.8 Hz, 1H, H-1), 4.30−4.10 (m, 3H, H-3, H-4 and H-5), 4.05 (t, J = 9.2 Hz, 1H, H-2), 3.00−2.73 (m, 2H, SCH_2CH_3), 1.29 (t, J = 7.5 Hz, 3H, SCH_2CH_3); ¹³C NMR (100 MHz,

 (C_5D_5N) δ 195.6, 153.9, 129.9 (2C), 126.8, 122.3 (2C), 87.0, 79.8, 78.4, 74.9, 74.2, 71.1 24.3, 15.4; IR (solid) 3347, 2926, 1489, 1283, 1198, 1101, 1072, 1018 cm⁻¹; MS (FAB) m/z (rel intensity) 361 (M + H+ , 5), 307 (30), 289 (15), 154 (100), 136 (70), 107 (20), 77 (15); HRMS (FAB) calcd for $C_{15}H_{21}O_6S_2$ (M + H⁺) 361.0774, found 361.0812. Anal. Calcd for $C_{15}H_{20}O_6S_2$: C, 49.98; H, 5.59. Found: C, 49.68; H, 5.87.

Methyl 6-O-(4-Tolyloxy)thiocarbonyl-β-D-glucopyranoside (6b, Entry 5, Table 3). Major product 6b: 315.3 mg, 91% yield. Minor product 6a: 20.1 mg, 6% yield. White solid: $R_f = 0.35$ (MeOH/CHCl₃, 10:90); mp 64–66 °C; $[\alpha]^{25}$ _D = −14.8 (c 1.05, CH₃OH); ¹H NMR (400 MHz, C_5D_5N C_5D_5N) δ 7.69 (br s, 1H, OH), 7.47 (br s, 1H, OH), 7.12 $(d, J = 8.5 \text{ Hz}, 2\text{H}, \text{PhH}), 7.07 (d, J = 8.5 \text{ Hz}, 2\text{H}, \text{PhH}), 5.42 (d, J =$ 11.5 Hz, 1H, H-6a), 5.22 (dd, J = 11.5, 5.2 Hz, 1H, H-6b), 5.01 (br s, 1H, OH), 4.76 (d, J = 7.8 Hz, 1H, H-1), 4.30−4.15 (m, 2H, H-3, H-4 and H-5), 4.05 (t, J = 8.3 Hz, 1H, H-2), 3.64 (s, 3H, OCH₃), 2.15 (s, 3H, 4-CH₃-Ph); ¹³C NMR (100 MHz, C₅D₅N) δ 195.9, 151.9, 136.4, 130.4 (2C), 122.0 (2C), 105.7, 78.2, 74.9, 74.7, 74.6, 71.2, 56.8, 20.6; IR (solid) 3368, 2918, 1506, 1285, 1219, 1190, 1034, 1015 cm[−]¹ ; MS (FAB) m/z (rel intensity) 345 (M + H⁺, 5), 307 (35), 289 (20), 165 (5), 154 (100), 136 (70), 107 (20), 77 (20); HRMS (FAB) calcd for $C_{15}H_{21}O_7S$ $(M + H^+)$ 345.1014, found 345.1017. Anal. Calcd for C₁₅H₂₀O₇S: C, 52.31; H, 5.85. Found: C, 52.22; H, 5.98.

Methyl 6-O-(2-Naphthoxy)thiocarbonyl-β-D-glucopyranoside (7b, Entry 6, Table 3). Major product 7b: 302.8 mg, 80% yield. Minor product 7a: 12.6 mg, 3% yield. White solid: $R_f = 0.35$ (MeOH/ CHCl₃, 10:90); mp 114–116 °C; $[\alpha]_{D}^{27} = -14.8$ (c 1.22, CH₃OH);
¹H NMR (400 MHz C D N) δ 7.89 (d I – 8.8 Hz 1H NaphH) ¹H NMR (400 MHz, C_5D_5N C_5D_5N) δ 7.89 (d, J = 8.8 Hz, 1H, NaphH), 7.90−7.80 (m, 2H, NaphH), 7.63 (d, J = 2.4 Hz, 1H, NaphH), 7.60− 7.40 (m, 2H, NaphH), 7.37 (dd, J = 8.8, 2.4 Hz, 1H, NaphH), 5.46 (d, $J = 11.0$ Hz, 1H, H-6a), 5.28 (dd, $J = 11.0$, 2.4 Hz, 1H, H-6b), 4.79 (d, J = 7.8 Hz, 1H, H-1), 4.35−4.15 (m, 3H, H-3, H-4 and H-5), 4.08 (t, J $= 8.3$ Hz, 1H, H-2), 3.66 (s, 3H, OCH₃); ¹³C NMR (100 MHz, (C_5D_5N) δ 195.8, 151.5, 134.1, 132.1, 129.9, 128.2, 128.1, 127.1, 126.5, 121.9, 119.4, 105.7, 78.2, 74.9, 74.8, 74.7, 71.2, 56.8; IR (solid) 3352, 2913, 1510, 1448, 1290, 1103, 1080, 1034 cm^{−1}; MS (FAB) *m/z* (rel $intensity)$ 381 $(M + H⁺, 20)$, 307 (25) , 289 (25) , 154 (100) , 136 (70) , 117 (25), 77 (30); HRMS (FAB) calcd for $C_{18}H_{21}O_7S$ $(M + H^+)$ 381.1003, found 381.1020.

Methyl 6-O-(4-Chlorophenoxy)thiocarbonyl-β-D-glucopyranoside (8b, Entry 7, Table 3). Major product 8b: 298.5 mg, 82% yield. Minor product 8a: 12.4 mg, 4% yield. White solid: $R_f = 0.30$ (MeOH/CHCl₃, 10:90); mp 136–13[8](#page-2-0) °C; $[\alpha]^{25}$ _D = -13.6 (c 1.08, CH₃OH); ¹H NMR $(400 \text{ MHz}, \text{C}_5\text{D}_5\text{N}) \delta$ 7.71 (br s, 1H, OH), 7.50 (br s, 2H, OH), 7.40 $(d, J = 9.0$ Hz, 2H, 4-Cl-PhH), 7.13 $(d, J = 9.0$ Hz, 2H, 4-Cl-PhH), 5.39 (dd, $J = 11.5$ Hz, 1H, H -6a), 5.19 (d, $J = 11.5$, 4.5 Hz, 1H, H -6b), 4.76 (d, J = 7.6 Hz, 1H, H-1), 4.30−4.22 (m, 1H, H-3), 4.22−4.13 (m, 2H, H-4 and H-5), 4.10–4.00 (m, 1H, H-2), 3.64 (s, 3H, OCH₃); ¹³C NMR (100 MHz, C_5D_5N) δ 195.3, 152.4, 131.9, 130.0 (2C), 124.1 (2C), 105.7, 79.7, 78.1, 74.9, 74.7, 71.2, 56.9; IR (solid) 3374, 2922, 1485, 1285, 1196, 1084, 1032, 1011 cm⁻¹; MS (FAB) m/z (rel $intensity)$ 365 $(M + H⁺, 10)$, 307 (25), 289 (20), 221 (10), 154 (100), 136 (75), 107 (30); HRMS (FAB) calcd for $C_{14}H_{18}ClO_7S (M + H^+)$ 365.0456, found 365.0488. Anal. Calcd for C₁₄H₁₇ClO₇S: C, 46.09; H, 4.70. Found: C, 46.01; H, 4.38.

Methyl 6-O-(4-Fluorophenoxy)thiocarbonyl-β-D-glucopyranoside (9b, Entry 8, Table 3). Major product 9b: 311.8 mg, 90% yield. Minor product 9a: 19.9 mg, 6% yield. White solid: $R_f = 0.30$ (MeOH/CHCl₃, 10:90); mp 57–59 °C; $[a]_D^{23} = -14.6$ (c 1.04, CH₃OH); ¹H NMR $(400 \text{ MHz}, \text{C}_5\text{D}_5\text{N}) \delta 8.58 \text{ (br s, 1H, OH)}, 7.16 \text{ (d, } J = 6.1 \text{ Hz}, 4\text{H}, 4-$ F-PhH), 5.40 (d, $J = 11.2$ Hz, 1H, H-6a), 5.20 (dd, $J = 11.2$, 5.3 Hz, 1H, H-6b), 4.76 (d, J = 7.6 Hz, 1H, H-1), 4.30−4.10 (m, 3H, H-3, H-4 and H-5), 4.05 (t, J = 8.3 Hz, 1H, H-2), 3.64 (s, 3H, OCH₃); ¹³C NMR (100 MHz, C_5D_5N) δ 195.7, 160.8 (d, J = 244.1 Hz), 150.1, 124.1 (d, J = 8.3 Hz, 2C), 116.5 (d, J = 24.0 Hz, 2C), 105.7, 78.1, 74.8 (2C), 74.6, 71.1, 56.8; IR (solid) 3362, 2903, 1501, 1283, 1213, 1184, 1082, 1011 cm⁻¹; MS (FAB) m/z (rel intensity) 349 (M + H⁺, 25), 317 (25), 307 (15), 237 (20), 154 (100), 137 (55), 85 (35), 77 (20); HRMS (FAB) calcd for $C_{14}H_{18}FO_7S$ $(M + H^+)$ 349.0752, found

349.0791. Anal. Calcd for C₁₄H₁₇FO₇S: C, 48.27; H, 4.92. Found: C, 48.04; H, 4.95.

Methyl 6-O-Benzoyl-β-D-glucopyranoside (11b, Entry 10, Table 3).^{11b,12} Major product 11b: 286.7 mg, 97% yield. Minor product 11a: 21.6 mg, 7% yield. White solid: $R_f = 0.47$ (MeOH/CHCl₃, 10:90); mp 129[−](#page-9-0)[13](#page-9-0)1 °C (lit.¹² mp 131−132 °C); [α]²⁵_D = −17.8 (c 1.03, [C](#page-2-0)H₃OH) [lit.¹² [α]²⁷_D = -24.2 (c 1.40, H₂O)]; ¹H NMR (400 MHz, (CD3)2CO) δ 8.1[0](#page-9-0)−8.00 (m, 2H, PhH), 7.70−7.60 (m, 1H, PhH), 7.60−7.45 ([m, 2](#page-9-0)H, PhH), 4.68 (dd, J = 11.7, 2.2 Hz, 1H, H-6a), 4.48 $(br s, 1H, OH)$, 4.46 (dd, J = 11.7, 5.9 Hz, 1H, H-6b), 4.37 (br s, 1H, OH), 4.31 (br s, 1H, OH), 4.25 (d, J = 7.8 Hz, 1H, H-1), 3.70−3.60 $(m, 1H, H-5)$, 3.55–3.40 $(m, 3H, H-3$ and H-4), 3.43 $(s, 3H, OCH₃)$, 3.23 (t, J = 7.2 Hz, 1H, H-2); ¹³C NMR (100 MHz, $(CD_3)_2CO$) δ 166.7, 133.9, 131.2, 130.2 (2C), 129.4 (2C), 105.1, 77.8, 74.8, 74.8, 71.4, 65.0, 56.6; IR (solid) 3399, 2878, 1713, 1271, 1072, 1045, 1016, 974 cm⁻¹; MS (FAB) *m/z* (rel intensity) 299 (M + H⁺, 20), 267 (10), 154 (100), 136 (70), 107 (25), 105 (20), 77 (20); HRMS (FAB) calcd for $C_{14}H_{19}O_7$ $(M + H^+)$ 299.1125, found 299.1154. Anal. Calcd for C₁₄H₁₈O₇: C, 56.37; H, 6.08. Found: C, 56.23; H, 6.06.

Methyl 6-O-Isobutyryl-β-D-glucopyranoside (12b, Entry 11, Table 3). Major product 12b: 243.4 mg, 92% yield. Minor product 12a: 1.0 mg, <1% yield. White solid: $R_f = 0.35$ (MeOH/CHCl₃, 10:90); mp 101−102 °C; [α]²⁶_D = −17.2 (α 1.01, CH₃OH); ¹H NMR (400 MHz, C_5D_5N C_5D_5N) δ 6.86 (br s, 3H, OH), 4.98 (d, J = 11.5 Hz, 1H, H-6a), 4.81 $(dd, J = 11.5, 5.9 Hz, 1H, H-6b), 4.70 (d, J = 7.8 Hz, 1H, H-1), 4.23 (t,$ J = 8.7 Hz, 1H, H-4), 4.10−3.95 (m, 3H, H-2, H-3 and H-5), 3.63 (s, 3H, OCH₃), 2.57 (sept, J = 6.8 Hz, 1H, OCOCH(CH₃)₂), 1.12 (d, J = 6.8 Hz, 3H, OCOCH $(CH_3)_2$), 1.10 (d, J = 6.8 Hz, 3H, OCOCH- $(CH_3)_2$); ¹³C NMR (100 MHz, CDCl₃) δ 177.7, 103.5, 76.1, 73.9, 73.4, 70.3, 63.5, 57.0, 33.9, 19.0, 18.9. IR (solid) 3412, 2907, 1717, 1192, 1161, 1098, 1049, 1003 cm⁻¹; MS (FAB) m/z (rel intensity) 265 (M + H⁺ , 15), 233 (20), 154 (100), 137 (60), 136 (65), 107 (15), 89 (15), 77 (10); HRMS (FAB) Calcd for $C_{11}H_{21}O_7$ $(M + H^+)$ 265.1282, found 265.1281. Anal. Calcd for C₁₁H₂₀O₇: C, 49.99; H, 7.63. Found: C, 49.72; H, 7.65.

Methyl 6-O-[3,5-Bis(trifluoromethyl)benzenesulfonyl]-β-D-glucopyranoside (13b, Entry 12, Table 3). Major product 13b: 472.9 mg, >99% yield. Minor product 13a: 1.9 mg, <1% yield. White solid: R_f = 0.31 (MeOH/CHCl₃, 10:90); mp 136–138 °C; $[\alpha]_{\text{D}}^{17} = -13.0$ (c 1.15, CH₃OH); ¹H NMR (400 M[Hz](#page-2-0), $(CD_3)_2CO$) δ 8.51 (s, 3H, 3,5- $CF_3\text{-}PhH$), 4.58 (dd, J = 11.0, 1.3 Hz, 1H, H-6a), 4.46 (dd, J = 11.0, 5.4 Hz, 1H, H-6b), 4.44 (br s, 1H, OH), 4.32 (br s, 1H, OH), 4.29 (br s, 1H, OH), 4.11 (d, J = 7.6 Hz, 1H, H-1), 3.60−3.45 (m, 1H, H-5), 3.40−3.20 (m, 2H, H-3 and H-4), 3.30 (s, 3H, OCH₃), 3.04 (t, J = 8.2 Hz, 1H, H-2); ¹³C NMR (100 MHz, $(CD_3)_2CO$) δ 140.0, 133.4 (q, J $= 34.8$ Hz, 2C), 129.3 (q, J = 3.3 Hz, 2C), 128.6 (sept, J = 3.3 Hz), 123.6 (q, J = 272.3 Hz, 2C), 104.9, 77.5, 74.4, 74.1, 72.2, 70.4, 56.7; ¹⁹F NMR (376 MHz, (CD₃)₂CO) δ –62.3 (m, 6F); IR (solid) 3376, 2922, 1607, 1466, 1362, 1281, 1179, 976 cm^{−1}; MS (FAB) *m/z* (rel intensity) 471 (M + H⁺, 15), 451 (10), 421 (55), 277 (30), 213 (60), 154 (100), 145 (40), 136 (95); HRMS (FAB) calcd for $C_{15}H_{17}F_6O_8S$ $(M + H⁺)$ 471.0543, found 471.0559. Anal. Calcd for $C_{15}H_{16}F_6O_8S$: C, 38.30; H, 3.43. Found: C, 38.01; H, 3.25.

Kinetics Experiment (Figure 1). To a solution of 1,2,3,4-Otetramethyl- α - or β -D-glucopyranoside¹³ (118.1 mg, 0.50 mmol) in THF (5 mL) in a vial was added pyridine (0.053 mL, 0.13 mmol) and isobutyryl chloride (0.50 mL, 1.1 m[m](#page-3-0)[ol\)](#page-9-0) at 0 °C. Then, after stirring vigorously for 5−240 min at 0 °C, the reaction mixture was quenched with saturated aqueous NH4Cl and extracted with ethyl acetate. The organic layer was washed with water and brine, dried over MgSO4, filtrated, and concentrated in vacuo. The residue was purified by $SiO₂$ column chromatography (n-hexane/ethyl acetate = 9:1−1:1) to give 1,2,3,4-O-tetramethyl-6-O-isobutyryl- α - or β-D-glucopyranoside 14a or 14b, respectively.

1,2,3,4-O-Tetramethyl-6-O-isobutyryl- α -D-glucopyranoside (14a, Figure 1). Colorless oil: $R_f = 0.52$ (EtOAc/n-hexane, 50:50); $[\alpha]^2$ $T_{\rm D}$ = +125.0 (c 1.07, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 4.81 (d, J = 3.7 Hz, 1H, H-1), 4.31 (dd, J = 11.8, 2.3 Hz, 1H, H-6a), 4.25 (dd, J = 11.8, 5[.0](#page-3-0) Hz, 1H, H-6b), 3.71 (ddd, J = 10.3, 5.0, 2.3 Hz, 1H, H-5), 3.63 (s, 3H, OCH₃), 3.53 (s, 3H, OCH₃), 3.53 (dd, J = 9.5, 8.9 Hz, 1H, H-3), 3.52 (s, 3H, OCH₃), 3.41 (s, 3H, OCH₃), 3.20 (dd, J = 9.5, 3.7 Hz, 1H, H-2), 3.08 (dd, $J = 10.3$, 8.9 Hz, 1H, H-4), 2.61 (sept, $J =$ 7.1 Hz, 1H, OCOCH $(CH_3)_2$), 1.20 (d, J = 7.1 Hz, 3H, OCOCH- $(CH_3)_2$), 1.19 (d, J = 7.1 Hz, 3H, OCOCH(CH₃)₂); ¹³C NMR (100 MHz, CDCl₃) δ 176.6, 97.2, 83.4, 81.6, 79.7, 68.6, 62.8, 60.8, 60.4, 58.9, 55.0, 33.8, 19.0, 18.8; IR (oil) 2934, 1734, 1470, 1387, 1190, 1155, 1098, 1043 cm⁻¹; MS (FAB) m/z (rel intensity) 307 (M + H⁺ , 20), 305 (20), 275 (70), 243 (90), 155 (80), 127 (50), 101 (100), 71 (85); HRMS (FAB) calcd for $C_{14}H_{27}O_7$ $(M + H^+)$ 307.1757, found 307.1757.

1,2,3,4-O-Tetramethyl-6-O-isobutyryl-β-D-glucopyranoside (14b, Figure 1). Colorless oil: $R_f = 0.58$ (EtOAc/n-hexane, 50:50); $[\alpha]_{D}^{26} =$ −13.1 (ϵ 1.10, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 4.34 (dd, J = 11.8, 2.2 Hz, 1H, H-6a), 4.21 (dd, J = 11.8, 5.6 Hz, 1H, H-6b), 4.16 (d, $J = 7.8$ [H](#page-3-0)z, 1H, H-1), 3.63 (s, 3H, OCH₃), 3.57 (s, 3H, OCH₃), 3.52 $(s, 3H, OCH_3)$, 3.52 $(s, 3H, OCH_3)$, 3.39 (ddd, J = 9.5, 5.6, 2.2 Hz, 1H, H -5), 3.19 (dd, J = 9.0, 8.8 Hz, 1H, H -3), 3.10 (dd, J = 9.5, 9.0 Hz, 1H, H-4), 3.00 (dd, $J = 8.8$, 7.8 Hz, 1H, H-2), 2.60 (sept, $J = 7.1$ Hz, 1H, OCOCH(CH₃)₂), 1.19 (d, J = 7.1 Hz, 3H, OCOCH(CH₃)₂), 1.18 $(d, J = 7.1 \text{ Hz}, 3H, OCOCH(CH_3)_2);$ ¹³C NMR (100 MHz, CDCl₃) δ 176.8, 104.1, 86.5, 83.6, 79.6, 72.8, 63.0, 60.8, 60.4, 60.4, 56.9, 33.9, 19.1, 18.8; IR (oil) 2936, 1736, 1470, 1387, 1192, 1150, 1109, 1076 cm⁻¹; MS (FAB) *m/z* (rel intensity) 307 (M + H⁺, 20), 305 (25), 275 (75), 243 (90), 155 (80), 127 (45), 101 (100), 71 (75); HRMS (FAB) calcd for $C_{14}H_{27}O_7$ $(M + H^+)$ 307.1757, found 307.1768.

General Procedure for the Chemo- and Regioselective Functionalization of Structurally Similar Carbohydrates Cata**lyzed by Bu₂SnCl₂ (Scheme 2 (a)).** After the mixture of methyl α -Dglucopyranoside (194.2 mg, 1.0 mmol), methyl β -D-arabinofuranoside (164.2 mg, 1.0 mmol), and dibutyltin dichloride (30.4 mg, 0.10 mmol) in THF (8 mL) was stirred in [a](#page-3-0) vial at room temperature for 10 min, tetrabutylammonium iodide (184.7 mg, 0.50 mmol), phenyl chlorothionoformate (0.175 mL, 1.3 mmol), and 1,2,2,6,6-pentamethylpiperidine (0.271 mL, 1.5 mmol) were added to the suspension at 20 \degree C. After being stirred vigorously for 6 h at 20 \degree C, the reaction mixture was quenched with saturated aqueous NH4Cl and extracted with ethyl acetate. The organic layer was washed with water and brine, dried over MgSO4, filtrated, and concentrated in vacuo (water bath temperature: <20 °C). The residue was purified by SiO_2 column chromatography (n-hexane/ethyl acetate = $3:1-0:1$) to give a mixture of methyl 2-O-phenoxythiocarbonyl-α-D-glucopyranoside 1a (320.4 mg, 97% yield) as a white solid.

Methyl 2-O-(3,5-Difluorobenzenesulfonyl)- α -D-glucopyranoside (15, Scheme 2 (a)).^{2j} Yield of 15: 352.5 mg, 95%; white solid; $R_f =$ 0.27 (MeOH/CHCl₃, 10:90); mp 38–40 °C; [α]²¹_D = +94.4 (c 1.13, CH₃OH); ¹H NMR [\(4](#page-9-0)00 MHz, $(CD_3)_2CO$) δ 7.70–7.60 (m, 2H, 3,5-F-PhH), 7.55[−](#page-3-0)7.45 (m, 1H, 3,5-F-PhH), 4.82 (d, J = 3.7 Hz, 1H, H-1), 4.67 (br s, 1H, OH), 4.48 (br s, 1H, OH), 4.28 (dd, J = 9.6, 3.7 Hz, 1H, H-2), 3.90−3.60 (m, 4H, H-3, H-6 and OH), 3.60−3.50 (m, 1H, H-5), 3.39 (t, J = 9.3 Hz, 1H, H-4), 3.34 (s, 3H, OCH₃); ¹³C NMR $(100 \text{ MHz}, (\text{CD}_3)_{2}\text{CO})$ δ 163.6 (d, J = 252.4 Hz), 163.5 (d, J = 252.4 Hz), 140.8 (t, J = 9.1 Hz), 112.6 (d, J = 19.9 Hz), 112.5 (d, J = 19.9 Hz), 110.2 (t, J = 25.7 Hz), 98.1, 82.2, 73.0, 71.7, 71.6, 62.1, 55.2; ¹⁹F NMR (376 MHz, $(CD_3)_2CO$) δ –106.4 to –106.5 (m, 2F). IR (solid) 3377, 2928, 1607, 1443, 1371, 1300, 1179, 962 cm⁻¹; MS (FAB) m/z (rel intensity) 371 (M + H+ , 10), 339 (20), 307 (20), 289 (15), 261 (5), 154 (100), 145 (30), 137 (90); HRMS (FAB) calcd for $C_{13}H_{17}F_2O_8S$ (M + H⁺) 371.0607, found 371.0629. Anal. Calcd for $C_{13}H_{16}F_2O_8S$: C, 42.16; H, 4.35. Found: C, 42.37; H, 4.11.

Methyl 3-O-Benzoyl- α -D-galactopyranoside (16, Scheme 2 (c)).^{11b} Major product 16: 297.9 mg, >99% yield. Minor product 11a: 56.7 mg, 19% yield. White solid: $R_f = 0.30$ (MeOH/CHCl₃, 10:[90\).](#page-9-0) Mp 55–57 °C. $[\alpha]^{27}$ _D = +190.1 (c 1.20, CH₃OH); ¹H NM[R](#page-3-0) $(400 \text{ MHz}, (\text{CD}_3)_2\text{CO}) \delta 8.08 \text{ (dd, } J = 8.4, 1.3 \text{ Hz, } 2\text{H, } \text{PhH}), 7.63 \text{ (tt, }$ $J = 7.4$, 1.3 Hz, 1H, PhH), 7.50 (t, $J = 7.7$ Hz, 2H, PhH), 5.22 (dd, $J =$ 10.3, 3.2 Hz, 1H, H-3), 4.80 (d, $J = 3.9$ Hz, 1H, H-1), 4.44 (br s, 1H, OH), 4.33–4.15 (m, 2H, H-2 and H-4), 3.89 (t, J = 5.7 Hz, 1H, H-5), 3.89 (br s, 1H, OH), 3.85−3.65 (m, 3H, H-6 and OH), 3.42 (s, 3H, OCH₃); ¹³C NMR (100 MHz, $(CD_3)_2$ CO) δ 166.7, 133.7, 131.5, 130.4 (2C), 129.1 (2C), 101.2, 75.5, 71.6, 68.5, 67.5, 62.2, 55.4; IR

(solid) 3385, 2932, 1697, 1315, 1273, 1119, 1053, 970 cm[−]¹ ; MS (FAB) m/z (rel intensity) 299 (M + H⁺, 5), 289 (15), 154 (100), 136 (90), 107 (15), 105 (10), 77 (10); HRMS (FAB) calcd for $C_{14}H_{19}O_7$ $(M + H⁺)$ 299.1125, found 299.1154.

Methyl 3-O-[3,5-Bis(trifluoromethyl)benzenesulfonyl]-α-D-galactopyranoside (17, Scheme 2 (c)). Major product 17: 425.3 mg, 90% yield. Minor product 13a: 22.4 mg, 5% yield. White solid: R_f = 0.41 (MeOH/CHCl₃, 10:90); mp 134–135 °C; $[\alpha]^{28}$ _D = +128.8 (c 1.[0](#page-3-0)1, CH₃OH); ¹H NMR (400 MHz, $(CD_3)_2CO$) δ 8.54 (s, 2H, 3,5- $CF_3\text{-}PhH$), 8.44 (s, 1H, 3,5-CF₃-PhH), 4.78 (dd, J = 10.3, 3.2 Hz, 1H, H-3), 4.69 (d, $J = 3.9$ Hz, 1H, H-1), 4.66 (d, $J = 5.1$ Hz, 1H, OH), 4.30 $(t, J = 4.0$ Hz, 1H, H-4), 4.03 (ddd, $J = 10.3, 3.9, 2.9$ Hz, 1H, H-2), 3.95−3.65 (m, 5H, H-5, H-6 and OH), 3.31 (s, 3H, OCH₃); ¹³C NMR (100 MHz, $(CD_3)_2CO$) δ 141.1, 133.0 (q, J = 34.8 Hz, 2C), 129.5 (q, $J = 3.3$ Hz, 2C), 128.2 (sept, $J = 3.3$ Hz), 123.6 (q, $J = 272.3$ Hz, 2C), 101.0, 85.2, 71.4, 69.4, 67.3, 61.9, 55.3; 19F NMR (376 MHz, $(CD_3)_2CO$) δ –62.3 (s, 6F); IR (solid) 3385, 2940, 1366, 1292, 1175, 1130, 1045, 962 cm⁻¹; MS (FAB) m/z (rel intensity) 471 (M + H⁺ , 30), 439 (75), 349 (40), 277 (20), 213 (20), 154 (85), 137 (100), 85 (40); HRMS (FAB) calcd for $C_{15}H_{17}F_6O_8S$ (M + H⁺) 471.0543, found 471.0547.

Methyl 3-O-Benzoyl-β-D-galactopyranoside (18, Scheme 2 (d)).^{11b,14} Major product 18: 298.4 mg, >99% yield. Minor product 11b: 26.9 mg, 9% yield. White solid: $R_f = 0.30$ (MeOH/CHCl₃, 10:[90\); m](#page-9-0)p 123−125 °C (lit.¹⁴ mp 120−121 °C); $\lceil \alpha \rceil^{27}$ $\lceil \alpha \rceil^{27}$ $\lceil \alpha \rceil^{27}$ _D = +52.4 (c 1.18, CH₃OH) [lit.¹⁴ [α]_D = +57.0 (c 1.00, C₂H₅OH)]; ¹H NMR (400 MHz, $(CD_3)_2CO$) δ 8.08 (d, J [=](#page-9-0) 7.6 Hz, 2H, PhH), 7.63 (t, J = 7.6 Hz, 1H, PhH), 7.51 (t, [J](#page-9-0) = 7.6 Hz, 2H, PhH), 4.99 (dd, J = 10.0, 2.7 Hz, 1H, H-3), 4.46 (br s, 1H, OH), 4.33 (d, J = 7.6 Hz, 1H, H-1), 4.31 (br s, 1H, OH), 4.24 (s, 1H, H-4), 3.97 (t, J = 8.8 Hz, 1H, H-2), 3.90−3.65 (m, 3H, H-6 and OH), 3.50 (s, 3H, OCH₃); ¹³C NMR (100 MHz, $(CD_3)_2CO$) δ 166.5, 133.7, 131.5, 130.4 (2C), 129.2 (2C), 105.6, 78.1, 75.7, 69.5, 67.7, 62.0, 56.7; IR (solid) 3393, 2936, 1705, 1450, 1273, 1117, 1067, 1026 cm⁻¹; MS (FAB) m/z (rel intensity) 299 (M + H+ , 30), 267 (55), 154 (100), 136 (65), 107 (25), 105 (95), 77 (25); HRMS (FAB) calcd for $C_{14}H_{19}O_7$ $(M + H^+)$ 299.1125, found 299.1153.

Methyl 3-O-[3,5-Bis(trifluoromethyl)benzenesulfonyl]-β-D-galactopyranoside (19, Scheme 2 (d)).^{2j} Major product 19: 458.8 mg, 97% yield. Minor product 13b: 45.4 mg, 10% yield. White solid: R_f = 0.34 (MeOH/CHCl₃, 10:90); mp [14](#page-9-0)4−146 °C; [α]²¹_D = +37.4 (c 1.10, CH₃OH); ¹H NMR (4[00](#page-3-0) MHz, $(CD_3)_2CO$) δ 8.53 (s, 2H, 3,5- $CF_3\text{-}PhH$), 8.45 (s, 1H, 3,5-CF₃-PhH), 4.64 (dd, J = 9.9 Hz, 1H, H-3), 4.58 (s, 1H, OH), 4.57 (s, 1H, OH), 4.30−4.20 (m, 1H, H-4), 4.14 (d, J = 7.6 Hz, 1H, H-1), 4.00−3.90 (m, 1H, H-5), 3.85−3.65 (m, 3H, H-2, H-6a and OH), 3.61 (dd, J = 11.7, 5.9 Hz, 1H, H-6b), 3.39 (s, 3H, OCH₃); ¹³C NMR (100 MHz, (CD₃)₂CO) δ 140.8, 132.8 (q, J = 34.8 Hz, 2C), 129.6 (q, J = 3.9 Hz, 2C), 128.2 (sept, J = 3.9 Hz), 123.6 (q, J $=$ 272.2 Hz, 2C), 105.0, 86.7, 75.1, 69.4, 68.7, 61.7, 56.8; ¹⁹F NMR $(376 \text{ MHz}, (\text{CD}_3)_2\text{CO})$ δ –62.4 (s, 6F); IR (solid) 3445, 2922, 1358, 1277, 1198, 1134, 1074, 961 cm⁻¹; MS (FAB) m/z (rel intensity) 471 (M + H+ , 15), 439 (35), 349 (20), 277 (15), 213 (15), 154 (100), 136 (85), 77 (35); HRMS (FAB) calcd for $C_{15}H_{17}F_6O_8S$ (M + H⁺) 471.0556, found 471.0562. Anal. Calcd for $C_{15}H_{16}F_6O_8S$: C, 38.30; H, 3.43. Found: C, 38.57; H, 3.03.

Methyl 3-O-Phenoxythiocarbonyl- α -D-mannopyranoside (20, Scheme 2 (b)).⁴ Major product 1a: 312.4 mg, 95% yield. Minor product 20: 34.7 mg, 10% yield. White solid: $R_f = 0.29$ (MeOH/ CHCl₃, 10:90); [m](#page-9-0)p 65–70 °C; $[\alpha]^{14}$ _D = +29.9 (c 1.40, CH₃OH); ¹H NMR (4[00](#page-3-0) MHz, C_5D_5N) δ 7.77 (br s, 3H, OH), 7.32 (t, J = 7.8 Hz, 2H, PhH), 7.22−7.18 (m, 1H, PhH), 6.97 (d, J = 8.3 Hz, 2H, PhH), 6.37−6.33 (m, 1H, H-3), 5.25 (br s, 1H, H-1), 5.14 (t, J = 9.6 Hz, 1H, H-4), 5.08 (d, $J = 1.5$ Hz, 1H, H-2), 4.56 (d, $J = 11.7$ Hz, 1H, H-6a), 4.46 (dd, J = 11.7, 5.3 Hz, 1H, H-6b), 4.39−4.33 (m, 1H, H-5), 3.44 (s, 3H, OCH₃); ¹³C NMR (100 MHz, C₅D₅N) δ 195.5, 153.9, 129.8 (2), 126.7, 122.3 (2), 102.7, 87.7, 75.5, 68.0, 65.4, 62.5, 54.6; IR (solid) 3354, 2928, 1489, 1271, 1190, 1128, 1057, 1016 cm[−]¹ ; MS (FAB) m/z (rel intensity) 331 (M + H⁺, 15), 307 (20), 289 (20), 154 (100), 136 (75), 107 (30), 77 (45), 65 (15); HRMS (FAB) calcd for

 $C_{14}H_{19}O_7S$ $(M + H^+)$ 331.0846, found 331.0869. Anal. Calcd for $C_{14}H_{18}O_7S$: C, 50.90; H, 5.49. Found: C, 50.63; H, 5.33.

Methyl 3-O-Benzoyl-α-D-mannopyranoside (21, Scheme 2
(b)).^{11b} Major product 11a: 296.2 mg, >99% yield. Minor product 21: 29.3 mg, 10% yield. White solid: $R_f = 0.31$ (MeOH/CHCl₃, 10:[90\);](#page-9-0) mp 33–34 °C (high moisture absorption); $[\alpha]^{27}$ _D = +17.5 ([c](#page-3-0) 0.90, CH₃OH); ¹H NMR (400 MHz, $(CD_3)_2CO$) δ 8.08 (d, J = 7.6 Hz, 2H, PhH), 7.63 (t, $J = 7.6$ Hz, 1H, PhH), 7.50 (t, $J = 7.6$ Hz, 2H, PhH), 5.23 (dd, J = 9.8, 3.2 Hz, 1H, H-3), 4.70 (s, 1H, H-1), 4.60– 4.45 (m, 2H, OH), 4.23−4.10 (m, 2H, H-2 and H-4), 3.88 (ddd, J = 11.5, 6.0, 2.8 Hz, 1H, H-6a), 3.77 (dd, J = 11.5, 6.3 Hz, 1H, H-6b), 3.70−3.55 (m, 2H, H-5, OH), 3.40 (s, 3H, OCH₃); ¹³C NMR (100 MHz, $(CD_3)_{2}CO$) δ 166.5, 133.7, 131.6, 130.5 (2C), 129.1 (2C), 102.3, 76.7, 74.4, 69.5, 65.8, 62.8, 54.8; IR (solid) 3385, 2932, 1697, 1315, 1273, 1119, 1053, 970 cm⁻¹; MS (FAB) m/z (rel intensity) 299 (M + H⁺ , 40), 267 (30), 154 (100), 136 (90), 107 (25), 105 (60), 77 (25); HRMS (FAB) calcd for $C_{14}H_{19}O_7$ $(M + H^+)$ 299.1125, found 299.1128.

Methyl 3-O-[3,5-Bis(trifluoromethyl)benzenesulfonyl]-α-D-mannopyranoside (22, Scheme 2 (b)).^{2j} Major product 13a: 450.9 mg, 96% yield. Minor product 22: 18.8 mg, 4% yield. White solid: $R_f = 0.36$ $(MeOH/CHCl₃, 10:90); mp 48–50 °C; [a]²¹_D = +15.6 (c 1.01,$ $(MeOH/CHCl₃, 10:90); mp 48–50 °C; [a]²¹_D = +15.6 (c 1.01,$ $(MeOH/CHCl₃, 10:90); mp 48–50 °C; [a]²¹_D = +15.6 (c 1.01,$ C[H](#page-3-0)₃OH); ¹H NMR (400 MHz, (CD₃)₂CO) δ 8.51 (s, 2H, 3,5-CF₃-PhH), 8.46 (s, 1H, 3,5-CF₃-PhH), 4.70 (d, J = 9.8, 3.2 Hz, 1H, H-3), 4.67 (d, J = 2.0 Hz, 1H, H-1), 4.09 (dd, J = 3.2, 2.0 Hz, 1H, H-2), 3.97 $(t, J = 9.8 \text{ Hz}, 1H, H-4)$, 3.75 (dd, $J = 12.0, 2.8 \text{ Hz}, 1H, H-6a)$, 3.65 (dd, J = 12.0, 4.9 Hz, 1H, H-6b), 3.60−3.20 (m, 4H, H-5 and OH), 3.33 (s, 3H, OCH₃); ¹³C NMR (100 MHz, (CD₃)₂CO) δ 140.9, 133.4 $(q, J = 34.8 \text{ Hz}, 2\text{C}), 129.5 (q, J = 3.3 \text{ Hz}, 2\text{C}), 128.3 \text{ (sept, } J = 3.3 \text{ Hz})$ Hz), 123.6 (q, J = 272.3 Hz, 2C), 102.1, 86.3, 74.3, 70.4, 65.4, 62.2, 54.9; ¹⁹F NMR (376 MHz, $(CD_3)_2CO$) δ –62.3 (s, 6F); IR (solid) 3389, 2940, 1362, 1279, 1179, 1132, 1057, 961 cm⁻¹; MS (FAB) m/z (rel intensity) 471 $(M + H^+, 10)$, 451 (5) , 439 (40) , 421 (45) , 277 (25), 213 (50), 154 (100), 136 (95); HRMS (FAB) calcd for $C_{15}H_{17}F_6O_8S$ (M + H⁺) 471.0543, found 471.0556. Anal. Calcd for $C_{15}H_{16}F_6O_8S$: C, 38.30; H, 3.43. Found: C, 38.01; H, 3.43.

General Procedure for the Chemo- and Regioselective Functionalization of Structurally Similar Carbohydrates Cata**lyzed by Me₂SnCl₂** (Scheme 2 (e)). After the mixture of methyl β -Dglucopyranoside (194.2 mg, 1.0 mmol), methyl β-D-galactopyranoside (194.2 mg, 1.0 mmol), and dimethyltin dichloride (22.0 mg, 0.10 mmol) in THF (9 mL) was stir[re](#page-3-0)d in a vial at room temperature for 10 min, 3,5-lutidine (0.0114 mL, 0.10 mmol) and 1,2,2,6,6-pentamethylpiperidine (0.271 mL, 1.5 mmol) were added to the suspension at 20 °C. Then, benzoyl chloride (0.128 mL, 1.1 mmol) in THF (2 mL) was flowed over 2 h at 20 °C. Then, after being stirred vigorously for 6 h at 20 °C, the reaction mixture was quenched with saturated aqueous NH4Cl and extracted with ethyl acetate. The organic layer was washed with water and brine, dried over MgSO₄, filtrated, and concentrated in vacuo. The residue was purified by $SiO₂$ column chromatography (nhexane/ethyl acetate = 3:1−0:1) to give a mixture of methyl 6-Obenzoyl-β-D-glucopyranoside 11b (284.4 mg, 96%) and methyl 3-Obenzoyl- β -D-galactopyranoside 18 (18.2 mg, 6%).

■ ASSOCIATED CONTENT

6 Supporting Information

Copies of spectra for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: muramatu@nagasaki-u.ac.jp.

Notes

The auth[ors declare no competing](mailto:muramatu@nagasaki-u.ac.jp) financial interest.

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■ REFERENCES

(1) For examples, see: (a) Feizi, T. Nature 1985, 314, 53−57. (b) Gabius, H.-J.; Gabius, S. Lectins and Cancer; Springer-Verlag: New York, 1991. (c) Crocker, P. R.; Feizi, T. Curr. Opin. Struct. Biol. 1996, 6, 679−691. (d) Kansas, G. S. Blood 1996, 88, 3259−3287. (e) Liedtke, S.; Geyer, H.; Wuhrer, M.; Geyer, R.; Frank, G.; Gerardy-Schahn, R.; Zahringer, U.; Schachner, M. Glycobiology 2001, 11, 373−384. (f) Brodesser, S.; Sawatzki, P.; Kolter, T. Eur. J. Org. Chem. 2003, 11, 2021−2034. (g) Almkvist, J.; Karlsson, A. Glycoconjugate J. 2004, 19, 575−581.

(2) For examples of the stereo- and regioselective synthesis, see: (a) Oshima, K.; Aoyama, Y. J. Am. Chem. Soc. 1999, 121, 2315−2316. (b) Herzner, H.; Reipen, T.; Schultz, M.; Kunz, H. Chem. Rev. 2000, 100, 4495−4538. (c) Martinelli, M. J.; Vaidyanathan, R.; Pawlak, J. M.; Nayyar, N. K.; Dhokte, U. P.; Doecke, C. W.; Zollars, L. M.; Moher, E. D.; Khau, V. V.; Košmrlj, B. J. Am. Chem. Soc. 2002, 124, 3578-3585. (d) Wang, C.-C.; Lee, J.-C.; Luo, S.-Y.; Kulkarni, S. S.; Huang, Y.-W.; Lee, C.-C.; Chang, K.-L.; Hung, S.-C. Nature 2007, 446, 896−899. (e) Kawabata, T.; Muramatsu, W.; Nishio, T.; Shibata, T.; Schedel, H. J. Am. Chem. Soc. 2007, 129, 12890−12895. (f) Lee, D.; Taylor, M. S. J. Am. Chem. Soc. 2011, 133, 3724−3727. (g) Hsu, C.-H.; Hung, S.-C.; Wu, C.-Y.; Wong, C.-H. Angew. Chem., Int. Ed. 2011, 50, 11872− 11923. (h) Jordan, P. A.; Miller, S. J. Angew. Chem., Int. Ed. 2012, 51, 2907−2911. (i) Lee, D.; Williamson, C. L.; Chan, L.; Taylor, M. S. J. Am. Chem. Soc. 2012, 134, 8260−8267. (j) Muramatsu, W. J. Org. Chem. 2012, 77, 8083−8091. (k) Lee, D.; Taylor, M. S. Synthesis 2012, 44, 3421−3431.

(3) For examples, see: (a) VanDraanen, N. A.; Koszalka, G. W. Nucleosides Nucleotides 1994, 13, 1679−1693. (b) Damkjaer, D. L.; Petersen, M.; Wengel, J. Nucleosides Nucleotides 1994, 13, 1801−1807. (c) García, J.; Fernández, S.; Ferrero, M.; Sanghvi, Y. S.; Gotor, V. Org. Lett. 2004, 6, 3759−3762. (d) García, J.; Díaz-Rodríguez, A.; Fernández, S.; Sanghvi, Y. S.; Ferrero, M.; Gotor, V. J. Org. Chem. 2006, 71, 9765−9771. (e) Maity, J.; Shakya, G.; Singh, S. K.; Ravikumar, V. T.; Parmar, V. S.; Prasad, A. K. J. Org. Chem. 2008, 73, 5629−5632. (f) Singh, S. K.; Sharma, V. K.; Bohra, K.; Olsen, C. E.; Prasad, A. K. J. Org. Chem. 2011, 76, 7556−7562.

(4) The use of a tetrabutylammonium iodide promotes the thiocarbonylation with activation of chlorothionoformates, see: Muramatsu, W.; Tanigawa, S.; Takemoto, Y.; Yoshimatsu, H.; Onomura, O. Chem.Eur. J. 2012, 18, 4850−4853.

(5) Vic, G.; Hastings, J. J.; Howarth, O. W.; Crout, D. H. G. Tetrahedron: Asymmetry 1996, 7, 709−720.

(6) Investigations of the coordination and effect of Sn reagent including 119Sn NMR analysis on the regioselectivity of dialkylstannylene acetal-mediated reactions have been reported. However, their ¹¹⁹Sn NMR analyses were carried out at high temperature or after dialkylstannylene acetals were prepared. For examples, see: (a) Grindley, T. B.; Thangarasa, R. Can. J. Chem. 1990, 68, 1007−1019. (b) Grindley, T. B.; Thangarasa, R. J. Am. Chem. Soc. 1990, 112, 1364−1373. (c) Grindley, R.; Thangarasa, T. B.; Bakshi, P. K.; Cameron, T. S. Can. J. Chem. 1992, 70, 197−204. (d) Kong, X.; Grindley, T. B. Can. J. Chem. 1994, 72, 2396−2404. (e) Kong, X.; Grindley, T. B. Can. J. Chem. 1994, 72, 2405−2415. (f) Reference 5a. In our case of the use of R_2SnCl_2 , on the other hand, we have not observed a peak of dialkylstannylene acetals so far under our best reaction conditions.

(7) Long-range stereoelectronic controlled reaction of carbohydrates in the presence of stereoisomers has been a few reported. For example, see: (a) Miljkovic, M.; Yeagley, D.; Deslongchamps, P.; Dory, Y. L. J. Org. Chem. 1997, 62, 7597−7604. (b) Magaud, D.; Dolmazon, R.; Anker, D.; Doutheau, A.; Dory, Y. L.; Deslongchamps, P. Org. Lett.

2000, 2, 2275−2277. (c) Zeng, Y.; Wang, Z.; Whitfield, D.; Huang, X. J. Org. Chem. 2008, 73, 7952−7962.

(8) We have not found reasonable data about the influences of anomeric effects so far in the screening of anomeric substituent (entries 1−4 of Table 3) due to the different solubilities of each carbohydrates in THF.

(9) All data of a minor product 20−22 in each reaction are shown in the Experimental Sectio[n](#page-2-0) and Supporting Information.

(10) Haque, M. E.; Kitauchi, T.; Kanemitsu, K.; Tsuda, Y. Chem. Pharm. Bull. 1987, 35, 1016−1029.

(1[1\) \(a\) Ho, W. M.; W](#page-4-0)ang, H. N. C. [Tetrahedron](#page-8-0) 1995, 51, 7373− 7388. (b) Tsuda, Y.; Haque, M. E.; Yoshimoto, K. Chem. Pharm. Bull. 1983, 31, 1612−1624.

(12) Bollenback, G. N.; Parrish, F. W. Carbohydr. Res. 1971, 17, 431− 438.

(13) (a) Haworth, W. N.; Hirst, E. L.; Miller, E. J. J. Chem. Soc. 1927, 2436−2443. (b) Bernet, B.; Vasella, A. Helv. Chim. Acta 1979, 62, 1990−2016. (c) Martín, A.; Salazar, J. A.; Suárez, E. J. Org. Chem. 1996, 61, 3999−4006. (d) Pinilla, I. M.; Martínez, M. B.; Galbis, J. A. Carbohydr. Res. 2003, 338, 549−555. (e) Boultadakis-Arapinis, M.; Lemoine, P.; Turcaud, S.; Micouin, L.; Lecourt, T. J. Am. Chem. Soc. 2010, 132, 15477−15479.

(14) Marra, A.; Sinay, P. Carbohydr. Res. 1990, 195, 303−308.